

1

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#### The fundamental ideas in chemistry

electron
 nucleus
 neutron
 each for 1 mark

2.

( m)

- B D
- (ii) D
  (b) A and B only one type of atom
  C and D more than one type of atom
  accept element for atom
  ignore the word 'mixaire'
  (chemically) bonded

accept (chemically) joined or similar idea of joined

- 3. B
   carbon monoxide
   1

   CO
   accept carbon oxide
   1

   do not credit carbon dioxide
   1
  - do not credit if any superscripts or subscripts used but accept C<sub>1</sub>O<sub>1</sub>, accept OC do not credit if obviously lower case
  - C water
    H<sub>2</sub>O accept hydrogen oxide
    do not accept hydrogen hydroxide
    do not credit if obviously lower case or if 2 not subscript
    do not accept HOH
    accept OHs
  - D ammonia do not accept ammonium 1
    NH3 do not credit if obviously lower case, or if 3 not subscript 1
    accept nitrogen hydride or hydrogen nitride
    do not accept hydrogen nitrate or nitrite
    allow H<sub>3</sub>N
- 4. correct reactants (i.e. sodium + water either way round)
  - correct products (i.e. sodium hydroxide + hydrogen, either way round)
- correct use of 'react'/'reaction'/reactants'/combine (not mixed'added/join) 3
  correct use of 'produces'/products'/gives/forms/makes/creates reactants
  correctly identified

  each for 1 mark
  - products correctly identified



1

# June 2014 C1 Mark Scheme

**Leigh Claire La Berge** 

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Los Angeles Times ,1974

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