

## CHEMISTRY

## FORM FIVE

FIRST TERM – 03.01.2007 to 09.03.2007 ( 10 weeks)

19.03.2007 to 25.05.2007 ( 10 weeks)

THEME  
LEARNING AREA::  
::INTERACTION BETWEEN CHEMICALS  
1. RATE OF REACTION

| No. of week/ Weeks | Learning Objectives               | Suggested Learning Activities                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                         | Learning Outcomes                                                                                                                                                                                                                                                                                                                                                                                                                                                                        | Thinking Skills                                                                                                                                                     | Scientific Skills                                                                                                                    | Scientific Attitudes And Noble Values                                                                                                                              | Remarks/ Vocabulary                                                                                                                                                                            |
|--------------------|-----------------------------------|-------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|---------------------------------------------------------------------------------------------------------------------------------------------------------------------|--------------------------------------------------------------------------------------------------------------------------------------|--------------------------------------------------------------------------------------------------------------------------------------------------------------------|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| 1                  | 1.1<br>Analysing rate of reaction | <p>Discusses :<br/>(a) the meaning of rate of reaction,<br/>(b) some examples of fast reactions,<br/>(c) some examples of slow reactions.</p> <p>Discusses to identify observable changes to reactants or products and its method of measurement in order to determine the rate of reaction.</p> <p>Carry out an activity involving a reaction between zinc and acid, and plot a graph to determine average rate of reaction and the rate of reaction at any given time.</p> <p>Carry out problem solving activities involving rates of reaction.</p> | <p>A student is able to:</p> <ul style="list-style-type: none"> <li>state what rate of reaction is,</li> <li>identify observable changes to reactants or products for determining rate of reaction,</li> <li>determine average rate of reaction,</li> <li>determine the rate of reaction at any given time from a graph,</li> <li>solve numerical problems involving average rate of reaction,</li> <li>solve numerical problem involving rate of reaction at any given time.</li> </ul> | <ul style="list-style-type: none"> <li>generating ideas</li> <li>relating</li> <li>making hypothesis</li> <li>making conclusion</li> <li>problem solving</li> </ul> | <ul style="list-style-type: none"> <li>Observing</li> <li>Hypothesising</li> <li>Communicating</li> <li>Interpreting data</li> </ul> | <ul style="list-style-type: none"> <li>Being honest and accurate in recording and validating data.</li> <li>Being cooperative</li> <li>Being systematic</li> </ul> | <p>reactant-<br/>bahan tindak balas</p> <p>product-<br/>hasil tindak balas</p> <p>rate of reaction-<br/>kadar tindak balas</p> <p>observable change-<br/>perubahan yang dapat diperhatikan</p> |

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| No. of week/ Weeks | Learning Objectives                                        | Suggested Learning Activities                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                | Learning Outcomes                                                                                                                                                                                                                                                                                                                                                                              | Thinking Skills                                                                                                                                                                                                                                            | Scientific Skills                                                                                                                 | Scientific Attitudes And Noble Values                                                                                                                                                                      | Remarks/ Vocabulary                                                |
|--------------------|------------------------------------------------------------|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|-----------------------------------------------------------------------------------------------------------------------------------|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|--------------------------------------------------------------------|
| 2                  | 1.2<br>Synthesising factors affecting the rate of reaction | <p>Discusses possible factors affecting the rate of reaction.</p> <p>Design and carry out activities to investigate factors affecting the rate of reaction, i.e. size of reactant, concentration, temperature and catalyst.</p> <p>Some suggested reactions:<br/>(a) a reaction between calcium carbonate, <math>\text{CaCO}_3</math>, and hydrochloric acid, <math>\text{HCl}</math>,<br/>(b) a reaction between sodium thiosulphate, <math>\text{Na}_2\text{S}_2\text{O}_3</math>, and sulphuric acid, <math>\text{H}_2\text{SO}_4</math>,<br/>(c) decomposition of hydrogen peroxide, <math>\text{H}_2\text{O}_2</math>, in the presence of a catalyst.</p> <p>Use computer simulations to investigate how the movement and collision of particles in a reaction are affected by temperature, size of reactant, pressure, concentration and catalyst.</p> <p>Collect and interpret data to explain factors affecting the rate of reaction in the following:<br/>(a) combustion of charcoal,<br/>(b) storing food in a</p> | <p>A student is able to:</p> <ul style="list-style-type: none"> <li>design experiment to investigate factors affecting the rate of reaction,</li> <li>give examples of reactions that are affected by size of reactant, concentration, temperature and catalyst,</li> <li>explain how each factor affects the rate of reaction,</li> <li>describe how factors affecting the rate of</li> </ul> | <ul style="list-style-type: none"> <li>generating ideas</li> <li>relating</li> <li>making hypothesis</li> <li>analyzing</li> <li>contributing</li> <li>making conclusion</li> <li>predicting</li> <li>making inference</li> <li>problem solving</li> </ul> | <ul style="list-style-type: none"> <li>experimenting</li> <li>predicting</li> <li>interpreting data</li> <li>inferring</li> </ul> | <ul style="list-style-type: none"> <li>Being honest and accurate in recording and validating data</li> <li>Being systematic</li> <li>Being cooperative</li> <li>Being confident and independent</li> </ul> | <p>catalyst -<br/>mangkin</p> <p>decomposition-<br/>penguraian</p> |

# Federal Scheme Of Work For Sschemistry

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