

Book 3: Example questions. All questions were translated from German. Keys are marked with an asterisk [*]. See text for details.

Question 1. Example of a kypnose-type question.

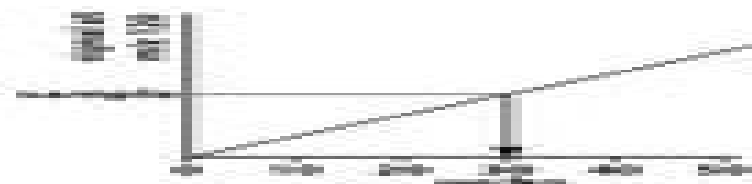
What increases the spectral resolution of an ICP-MS instrument?

- | | | |
|-----------------------|-----------------------|---|
| True | False | |
| ☐ | ☐ | Scatter plate in a sector field instrument |
| ☐ | ☐ | Using a dual mode detector with analog and pulse counting |
| ☐ | ☐ | Detection of barium at m/z 138 instead of 137 |
| ☐ | ☐ | Increase the length of the quadrupole for an ICP-QMS |

Question 2. Silver has a mass of 107.87 g/mol. At which m/z do you detect silver with ICP-MS?

- | | |
|----|-----|
| A | 106 |
| B* | 107 |
| C | 108 |
| D* | 109 |

Question 3. You quantify tin in a sample with ICP-MS and external calibration at m/z 120. The isotope abundance of ^{118}Sn is about 30%. The calibration for the signal at m/z 120 indicates a mass fraction of 30 $\mu\text{g}/\text{kg}$ [see figure below]. What is the mass fraction of tin in the measurement aliquot?



- | | |
|----|----------------------------|
| A | 90 $\mu\text{g}/\text{kg}$ |
| B | 60 $\mu\text{g}/\text{kg}$ |
| C* | 30 $\mu\text{g}/\text{kg}$ |
| D | 10 $\mu\text{g}/\text{kg}$ |

Question 4. Chocolate was digested in a closed vial under pressure using nitric acid and hydrogen peroxide using a microwave system. What constitutes the main matrix of the analyte cadmium in the prepared solution after the digestion of chocolate?

- | | |
|----|-------------------|
| A | cadm., sugar, fat |
| B* | water |
| C | cadmium ions |
| D | no matrix |

Question 5. How do you prepare a sample of selenium to differentiate between and quantify organic mercury species?

- | | |
|----|---|
| A | microwave digestion using HCl and H_2O_2 |
| B | microwave digestion using HNO_3 and H_2O_2 |
| C | microwave digestion using HNO_3 |
| D* | no digestion |

Multiple Choice Question For Instrumental Chemistry

Arshad Iqbal



Multiple Choice Question For Instrumental Chemistry:

Multiple Choice Question For Instrumental Chemistry Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Multiple Choice Question For Instrumental Chemistry**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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