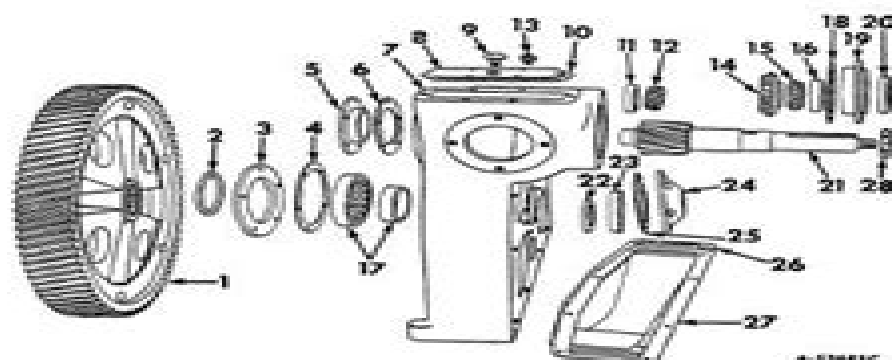


MAIN DRIVE



1	858 810 R81	GEAR ASSY. MAIN DRIVE
	105 814	RIVET, 1/2 X 2-1/4 BUTT HD -B-
	271 789	SCREW, 3/4-18 X 3-1/2 2ND HEX-HD CAP
	121 048	WASHER, 3/4 C-2 WED LOCK
2	863 867 R81	SEAL, CENTER BRG OIL
3	863 143 R1	RETAINER, PLUNGER CRANK BEARING
		SCREW, 5/8-18 X 1-1/4 HEX-HD CAP -B-
		WASHER, 3/8 C-2 WED LOCK -B-
4	863 401 R1	GASKET, PLUNGER CRANK BEARING RETAINER
5	863 168 R1	CASE, PINION SHAFT LEFT BRG
	179 841	SCREW, 3/8-18 X 1-1/4 HEX-HD WASH
		-B-
		WASHER, 3/8 WED LOCK -B-
6	863 403 R1	SKIN, PINION SHAFT LEFT BRG .005 -3-
	863 404 R1	SKIN, PINION SHAFT LEFT BRG .007 -3-
	863 427 R1	SKIN, PINION SHAFT LEFT BRG .020 -3-
7	863 180 R2	CASE, MAIN DRIVE GEAR
	438 189	FLDG, 1/2-14 AUTOMOTIVE SQ-HD PIPE
	180 178	SCREW, 1/2-13 X 1-1/4 HEX-HD CAP
		-10-
	120 384	WASHER, 1/2 C-2 WED LOCK -10-
8	863 805 R1	COVER, GEAR CASE
	18 763 R1	BOLT, 5/16-18 X 1/2 C-2 HEX-HD WASH
		-B-
	120 214	WASHER, 5/16 C-2 WED LOCK -B-
9	153 127 H1	BREATHER, MAIN DRIVE GEAR CASE
		-REPLACES 863 002 R81-
10	863 406 R1	GASKET, GEAR CASE COVER
11	120 503 H	BEARING, FLYWHEEL & PINION SHAFT CUP
12	62 923 R81	BEARING ASSY. FLYWHEEL & PINION SHAFT CONE
13	118 922	BUSHING, 3/8 X 1/8 REDUCING PIPE
14	863 229 R1	PINION, PACKER DRIVE BEVEL
	112 128	KEY, 1/4 X 7/8 WOODRUFF A

EXPLoded view of main drive gear case.

MAIN DRIVE -CONTINUED

15	858 458 R81	BEARING ASSY. PINION SHAFT RIGHT CONE
16	858 459 R1	BEARING, PINION SHAFT RIGHT CUP
17	863 565 R81	BEARING ASSY. CRANK CENTER
18	863 440 R1	SKIN, PINION SHAFT RIGHT BRG CASE .005
		-AS REQUIRED-
18	863 441 R1	SKIN, PINION SHAFT RIGHT BRG CASE .007
		-3-
18	863 442 R1	SKIN, PINION SHAFT RIGHT BRG CASE .020
		-3-
19	863 148 R1	CASE, PINION SHAFT RIGHT BRG
		SCREW, 3/8-18 X 1-1/4 HEX-HD CAP -B-
		WASHER, 3/8 WED LOCK -B-
20	11 848 F	SEAL, PINION SHAFT OIL
21	863 541 R1	PINION, MAIN DRIVE
	80	KEY, 1/2 X 3 SQ PARALLEL
	17 233 R1	PIN, 1/8 X 2-1/2 C-2 DOT.
22	863 344 R1	SPACER, PLUNGER CRANK BEARING
23	ST 270 A	BEARING ASSY. CRANK RIGHT
24	863 168 R1	RETAINER, PLUNGER CRANK RIGHT BRG
		SCREW, 5/8-18 X 1-1/4 HEX-HD CAP -B-
		WASHER, 3/8 WED LOCK -B-
25	863 447 R1	GASKET, PLUNGER CRANK BEARING
26	863 148 R1	PAN, MAIN DRIVE GEAR CASE
	438 189	FLDG, 1/2-14 AUTOMOTIVE SQ-HD PIPE
	180 081	SCREW, 5/16-18 X 1-1/4 C-2 HEX-HD
		CAP -B-
	180 112	SCREW, 5/16-18 X 2-3/4 2ND HEX-HD
		CAP -B-
	120 214	WASHER, 5/16 C-2 WED LOCK -10-
27	863 486 R1	GASKET, GEAR CASE PAN
28	870 372 R1	NUT, FLYWHEEL SLOTTED JAW

EXPLoded view of main drive gear case.

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web we can balance it for charge by adding two electrons to the right side of the equation so that the net charge on each side is 0 oxidation $\text{Ni} \rightarrow \text{Ni}^{2+} + 2\text{e}^-$ now that the oxidation half reaction is balanced it tells us that two electrons are produced for every atom of nickel oxidized but where do those electrons go

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web balancing redox reactions in acidic solution problems 1 10 problem 1 $\text{Cr}_2\text{O}_7^{2-} + \text{Fe}^{2+} + \text{H}^+ \rightarrow \text{Cr}^{3+} + \text{Fe}^{3+} + \text{H}_2\text{O}$ solution 1 balanced half reactions $6\text{e}^- + 14\text{H}^+ + \text{Cr}_2\text{O}_7^{2-} \rightarrow 2\text{Cr}^{3+} + 7\text{H}_2\text{O}$ $\text{Fe}^{2+} \rightarrow \text{Fe}^{3+} + \text{e}^-$ 2 equalize the electrons $6\text{e}^- + 14\text{H}^+ + \text{Cr}_2\text{O}_7^{2-} \rightarrow 2\text{Cr}^{3+} + 7\text{H}_2\text{O}$ $6\text{Fe}^{2+} \rightarrow 6\text{Fe}^{3+} + 6\text{e}^-$ multiplied by a factor of 6

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web when balancing equations for redox reactions occurring in acidic solution it is often necessary to add H^+ ions or the H_2O pair to fully balance the equation in this video we ll walk through this process for the reaction between dichromate $\text{Cr}_2\text{O}_7^{2-}$ and chloride Cl^- ions in acidic solution created by jay

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web balance each half reaction in basic solution 4 5 6 7 $\text{Cr}_2\text{O}_7^{2-} + 2\text{NO}_2 + \text{SO}_4^{2-} + \text{MnO}_2 + \text{Cr}^{3+} + \text{NO}_3^- + \text{SO}_2 + \text{Mn}_2\text{O}_3$ balance each redox reaction in acid solution using the half reaction method 8 $\text{H}_2\text{O}_2 + \text{Cr}_2\text{O}_7^{2-} + 2\text{H}^+ \rightarrow \text{TeO}_3 + 2\text{N}_2\text{O}_4 + \text{O}_2 + \text{Cr}^{3+} + \text{Te} + 10\text{H}_2\text{O}$ 11 $\text{PbO}_2 + \text{IO}_3^- + \text{H}^+ \rightarrow \text{Pb}^{2+} + \text{IO}_2$ 12 $\text{AsO}_3^{3-} + \text{NO}_3^- + \text{H}^+ \rightarrow \text{AsO}_4^{3-} + \text{H}_2\text{AsO}_4^-$

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web practice exercises balance the following equations of redox reactions assign oxidation numbers to all elements in the reaction separate the redox reaction into two half reactions balance the atoms in each half reaction add the two half reactions together and cancel out common terms

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web aug 29 2023 some points to remember when balancing redox reactions the equation is separated into two half equations one for oxidation and one for reduction the equation is balanced by adjusting coefficients and adding H^+ H_2O OH^- and e^- in this order balance the atoms in the equation apart from o and h

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web in the redox equation below that occurs in our stomach which of the following pairs identifies the reducing and oxidizing agents respectively $\text{C}_6\text{H}_8\text{O}_6 + 2\text{HNO}_2 \rightarrow \text{C}_6\text{H}_6\text{O}_6 + 2\text{H}_2\text{O} + 2\text{NO}$ choose 1 answer $\text{C}_6\text{H}_8\text{O}_6$ and NO_2 a $\text{C}_6\text{H}_8\text{O}_6$ and NO_2 no 2 and $\text{C}_6\text{H}_8\text{O}_6$ b NO_2 and $\text{C}_6\text{H}_8\text{O}_6$ h and NO_2 c h and