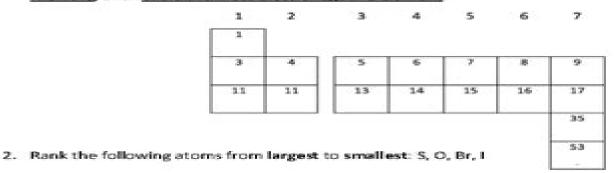
General Chemistry Review (Chapters 1 - 3)

 The first two rows of the periodic table, excluding noble gases and including larger halogens, is sufficient for most organic chemistry applications. Try to <u>fill in the periodic table below from</u> memory, then circle the most electronegative element.



- How many and what type of orbitals are in the second electron shell? Draw the shape and orientation of each.
- Identify the group of elements that corresponds to the following generalized electron configuration: [noble gas] ns²np³
- Draw Lewis structures for the following molecular formulas. Don't forget lone pairs. (Hint: these molecules contain only SINGLE bonds.)

 H_2O_2 CNH₅ C₃H₆

Draw Lewis structures for the following molecular formulas. Include Ione pairs. (Hint: these molecules contain double and/or triple bonds.)

CO₂ C₂H₄O HCN

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William Collins

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