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Covalent Bonding

To gain a stable electron arrangement, non metal elements share their outer electrons. This is called _____ bonding.

To be stable, an atom needs to have **8** electrons in its outer electron shell. Neon is a noble gas.



The noble gases are the only elements that are _____ on their own.

non-metals

are on the _____ hand side of the periodic table.

these are called "dot and cross" diagrams

each **X** and each **o** represents an electron

example



the fluorine atoms are unstable on their own

when they share electrons, each atom has _____ outer electrons = making them stable



covalent substances can be

covalent molecular or covalent network

Glencoe Chemistry Guided Notes Chapter 9 Covalent Bonding

Sabine Zange



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