

A crystalline solid is formed when 2 or more atoms exchange valence electrons to increase stability.

Ionic bond is formed when valence electrons are transferred from one atom to another.

cations = atoms lose electrons to become [+] charged ions.

atoms gain electrons to become [-] charged.

In ionic compounds that ions are arranged in a crystal lattice. Strong forces hold the ions together.

Properties of ionic compounds:

- = high melting and boiling points
- = hard & not easily crushed
- = conduct electricity when melted or dissolved because the ions are free to move

covalent bond = electrons are shared. Forming molecules

Covalent compounds have weak forces holding the molecules together.

Properties of covalent compounds:

- = low melting and boiling points
- = Many covalent compounds are volatile liquids or gases.
- = soft, easier to crush
- = are not conductors of electricity
- = electronegativity = property that tells how strong an

Episode 5intro To Bonding

O García



Episode 5intro To Bonding:

Unveiling the Power of Verbal Art: An Emotional Sojourn through **Episode 5intro To Bonding**

In some sort of inundated with screens and the cacophony of immediate communication, the profound energy and psychological resonance of verbal art often fade into obscurity, eclipsed by the regular onslaught of sound and distractions. Yet, situated within the lyrical pages of **Episode 5intro To Bonding**, a fascinating work of literary beauty that impulses with fresh thoughts, lies an memorable journey waiting to be embarked upon. Penned by a virtuoso wordsmith, that mesmerizing opus guides readers on an emotional odyssey, lightly exposing the latent possible and profound impact embedded within the intricate web of language. Within the heart-wrenching expanse of the evocative evaluation, we will embark upon an introspective exploration of the book is central styles, dissect its charming writing model, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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