

Chemistry EOC Review Packet

Matter and Change: Pages 34-50

Procedural:

- Chemical change
- Chemical property
- Physical change
- Physical property
- Homogeneous mixture
- Heterogeneous mixture
- Compound
- Element
- Atom
- Isotope
- Atom
- Law of Conservation of Mass

Concept Questions:

- Write the definition of each: (1) a homogeneous mixture and (2) a heterogeneous mixture.

Homogeneous - fluid, homogeneous and same appearance. **Heterogeneous** - heterogeneous, fluid, fluid

- Fill in the blank with whether the mixture should be homogeneous or heterogeneous.

Mixture of Matter	Homogeneous	Heterogeneous
Salad	no	yes
Spaghetti	no	yes
Gas	yes	no

- Which picture represents each state of matter?



Solid



Liquid



Gas

- What properties distinguish plasmas?

Solid - plasma is made of ionized gas particles

- What properties distinguish plasmas?

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LP Steffe

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'LP Steffe' text.

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