

EXPERIMENTAL PROCEDURES

1. **Introduction** 2. **Background** 3. **Methodology** 4. **Results** 5. **Discussion** 6. **Conclusion** 7. **References**

10. All of the following are among factors of the environment except:
 - a. Air
 - b. Temperature
 - c. Daylight
 - d. Soil
11. One source, plants, animals and fungi, contribute to:
 - a. Matter
 - b. Community
 - c. Habitat
 - d. Ecosystem
12. Which (or two) are examples of sufficient (or similar) requirements being in a system when?
 - a. Ecosystem
 - b. Ecosystems
 - c. Ecosystem
 - d. Community
13. Which (or two) are examples of sufficient (or similar) requirements being in a system when?
 - a. Ecosystem
 - b. Ecosystem
 - c. Ecosystem
 - d. Ecosystem
14. Which (or two) are examples of sufficient (or similar) requirements being in a system when?
 - a. Ecosystem
 - b. Ecosystem
 - c. Ecosystem
 - d. Ecosystem
15. Which (or two) are examples of sufficient (or similar) requirements being in a system when?
 - a. Ecosystem
 - b. Ecosystem
 - c. Ecosystem
 - d. Ecosystem
16. Which (or two) are examples of sufficient (or similar) requirements being in a system when?
 - a. Ecosystem
 - b. Ecosystem
 - c. Ecosystem
 - d. Ecosystem
17. Which (or two) are examples of sufficient (or similar) requirements being in a system when?
 - a. Ecosystem
 - b. Ecosystem
 - c. Ecosystem
 - d. Ecosystem
18. Which (or two) are examples of sufficient (or similar) requirements being in a system when?
 - a. Ecosystem
 - b. Ecosystem
 - c. Ecosystem
 - d. Ecosystem
19. Which (or two) are examples of sufficient (or similar) requirements being in a system when?
 - a. Ecosystem
 - b. Ecosystem
 - c. Ecosystem
 - d. Ecosystem
20. Which (or two) are examples of sufficient (or similar) requirements being in a system when?
 - a. Ecosystem
 - b. Ecosystem
 - c. Ecosystem
 - d. Ecosystem

Environmental Science Multiple Choice Questions 2008

CH Cherryholmes



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web write the oxidation and reduction reactions for each redox reaction the first one is done for you 23 $\text{Fe}^{2+} + \text{Co} \rightarrow \text{Co}^{2+} + \text{Fe}$
oxidation $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^{-}$ reduction $\text{Fe}^{2+} + 2\text{e}^{-} \rightarrow \text{Fe}$ 24 3 $\text{Ag} + \text{Ni}^{3+} \rightarrow \text{Ni} + 3\text{Ag}^{+}$ oxidation reduction 25 $\text{Cu}^{2+} + \text{Pb} \rightarrow \text{Pb}^{2+} + \text{Cu}$ oxidation reduction 26 $\text{O}_2 + 2\text{H}_2 \rightarrow 2\text{H}_2\text{O}$

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