

CHEM 8212 Advanced Analytical Chemistry
Midterm Exam
March 24, 2020

Answer the following questions about each topic:

1. (30 pts) **Metal nanoparticles:**

- a) (10 pts) Explain localized surface plasmon resonance (LSPR).
- b) (10 pts) Why do gold nanorods have two absorption peaks?
- c) (10 pts) Give one method to make gold nanorods and outline the major steps.

a) LSPR is the collective oscillation of free electrons (conduction band electrons) of a metal that is in resonance with the incident light and confined to a finite volume such as a nanoparticle.

b) Gold nanorods have two absorption peaks because the free electrons can collectively oscillate in two directions, one along with the long axis of the rod resulting in longitudinal absorption peak, and the other along with the short axis resulting in transverse absorption peak.

c) The most common method to make gold nanorods is Seed-Mediated Growth method. The major steps are:
Step 1: Preparation of the seed. In this step, ice-cold NaBH_4 reduces HAuCl_4 solution to form small gold nanoparticles of 2-5 nm.

Step 2: Growth of gold nanorods. In a growth solution containing HAuCl_4 , AgNO_3 , and CTAB, ascorbic acid is added to reduce the gold salt to Au^0 precursors. Then the seed solution is introduced to initiate the anisotropic growth of the gold seed into rod shape.

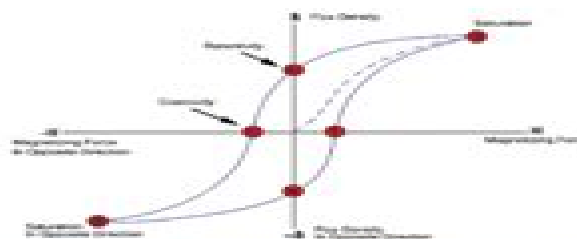
2. (30 points) **Magnetic nanoparticles:**

- a) (20 pts) Schematically compare the differences of superparamagnetic and ferromagnetic properties.
- b) (10 pts) Design a nanostructure that can be used for both magnetic resonance imaging and dark field light scattering imaging.

a) **Superparamagnetic**



Ferromagnetic



b) Iron oxide-gold core-shell nanoparticles. Due to the magnetic properties of the iron oxide core, the nanoparticles can be used for magnetic resonance imaging. Due to the optical properties of the gold shell, the nanoparticles can be used for dark field light scattering imaging.

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Christian Drosten

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1. Understanding the eBook Midterm Exam Answer Chem 24
 - The Rise of Digital Reading Midterm Exam Answer Chem 24
 - Advantages of eBooks Over Traditional Books
2. Identifying Midterm Exam Answer Chem 24
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Midterm Exam Answer Chem 24
 - User-Friendly Interface
4. Exploring eBook Recommendations from Midterm Exam Answer Chem 24
 - Personalized Recommendations
 - Midterm Exam Answer Chem 24 User Reviews and Ratings
 - Midterm Exam Answer Chem 24 and Bestseller Lists
5. Accessing Midterm Exam Answer Chem 24 Free and Paid eBooks
 - Midterm Exam Answer Chem 24 Public Domain eBooks
 - Midterm Exam Answer Chem 24 eBook Subscription Services
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- ePub, PDF, MOBI, and More
 - Midterm Exam Answer Chem 24 Compatibility with Devices
 - Midterm Exam Answer Chem 24 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Midterm Exam Answer Chem 24
 - Highlighting and Note-Taking Midterm Exam Answer Chem 24
 - Interactive Elements Midterm Exam Answer Chem 24
 8. Staying Engaged with Midterm Exam Answer Chem 24
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Midterm Exam Answer Chem 24
 9. Balancing eBooks and Physical Books Midterm Exam Answer Chem 24
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Midterm Exam Answer Chem 24
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Midterm Exam Answer Chem 24
 - Setting Reading Goals Midterm Exam Answer Chem 24
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Midterm Exam Answer Chem 24
 - Fact-Checking eBook Content of Midterm Exam Answer Chem 24
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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