Anion	Test	Result
Carbonate (CO ₃ 2-)	Add dilute acid (e.g. dilute hydrochloric acid)	Effervescence, CO ₂ produced (turns limewater milky)
Sulfite (SO₃²·)	Add dilute acid (e.g. dilute hydrochloric acid), warm	SO ₂ produced (turns aq. potassium manganate (VII) from purple to colourless
Choride (Cl) (in solution)	Acidify, then add aq. silver nitrate [Ag*] Acidify, then add aq. lead nitrate [Pb2*]	White ppt. of AgCl Ag* (aq) + Cl* (aq) → AgCl (s) White ppt. of PbCl ₂
Bromide (Br) [in solution]	Acidify, then add aq. silver nitrate	Cream ppt. of AgBr
lodide (I ⁻) [in solution]	Acidify, then add aq. silver nitrate (or lead nitrate)	Yellow ppt. of Agl (or Pbl ₂)
Sulfate (SO ₄ 2-) [in solution]	Acidify, then add aq. barium nitrate (or barium chloride) [Ba ²⁺]	White ppt. of BaSO₄ Ba ²⁺ + SO₄ ²⁻ → BaSO₄
Nitrate (NO ₃)	Add aq. sodium hydroxide, then little aluminium (foil) and warm carefully	Ammonia produced (turns damp red litmus paper or universal indicator paper blue)

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