

**PUNJAB PUBLIC SERVICE COMMISSION
COMBINED COMPETITIVE EXAMINATION
FOR RECRUITMENT TO THE POSTS OF
PROVINCIAL MANAGEMENT SERVICE, ETC.**

CHEMISTRY (OPTIONAL) PAPER-I

TIME ALLOWED: 3 HOURS

MAXIMUM MARKS: 100

Note: Attempt any FIVE questions only. Write answers in English.

- Q.1:** (a) Using a diagram, explain the variation of molecular energy with internuclear separation. (10)
- (b) What is the primary assumption of the VSEPR model? Explain the difference between square-pyramidal and trigonal bi-pyramidal arrangement in terms of VSEPR. Predict the shape of an SeF_4 molecule. (10)
- Q.2:** (a) What are the mineral sources for copper? Describe in detail the information which can be obtained for 3d metals using a FROST diagram. (10)
- (b) (i) On the basis of trends in the 3d series, suggest possible M^{2+} aqua ions for use as reducing agents, and write a balanced equation for the reaction of one of these ions with O_2 in acidic solution. (5)
- (ii) Comment on the 2+ oxidation states of 4d and 5d elements. (5)
- Q.3:** (a) Explain with the help of a flow sheet diagram, the electrolytic process of caustic soda manufacturing. (10)
- (b) Give a detailed comparison between dry and wet processing of cement manufacturing. (10)
- Q.4:** (a) Discuss the physical origin of quantization energy for a particle confined to moving inside a one dimensional box or on a ring. (10)
- (b) Use EHT to calculate the energies of the three molecular orbitals in an H_3^+ molecule ion treated as an equilateral triangle of side $R = 3.25a_0$ (170 pm). (10)
- Q.5:** (a) Calculate the work done when 50g of iron reacts with HCl to produce hydrogen gas in: (5)
- (i) a closed vessel of fixed volume, (ii) an open beaker at $25^\circ C$.
- (b) The molar constant pressure heat capacity of a certain solid at 10K is $0.43 J K^{-1} mol^{-1}$. What is its molar entropy at that temperature? (10)
- Q.6:** (a) Suggest a calibration procedure for a Na^+ selective electrode, in which the values of E_{ap} , β and k_{xy} are measured experimentally. Assume that K^+ is the only interfering species. (10)
- (b) Distinguish between galvanic and electrolytic cells. (10)
- Q.7:** (a) Define Nuclear Fission and Nuclear Fusion processes. Discuss their comparative features. (10)
- (b) Outline a method for preparation of CAN pellets. Distinguish between calcium superphosphate and Triple Superphosphate. (10)

June 2012 66ch01 Chemistry Past Paper

F Rizvi



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June 2012 66ch01 Chemistry Past Paper: Bestsellers in 2023 The year 2023 has witnessed a noteworthy surge in literary brilliance, with numerous engrossing novels captivating the hearts of readers worldwide. Lets delve into the realm of top-selling books, exploring the captivating narratives that have charmed audiences this year. June 2012 66ch01 Chemistry Past Paper : Colleen Hoovers "It Ends with Us" This heartfelt tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover skillfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can succeed. Uncover the Best : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This intriguing historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids absorbing storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Discover the Magic : Delia Owens "Where the Crawdads Sing" This captivating coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens spins a tale of resilience, survival, and the transformative power of nature, entrancing readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of captivating stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a masterful and suspenseful novel that will keep you wondering until the very end. The novel is a warning tale about the dangers of obsession and the power of evil.

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