

Chapter Test B

Part B: Matching and Completion

Matching Match the items of the column on the left with the items of the column on the right. Write the letter of the item on the right next to the item on the left.

- | | |
|---|--------------------|
| _____ 1. rate of two reactions with different activation energies in a given medium | a. chemical energy |
| _____ 2. decomposition reaction of the reactants in a given medium | b. reaction time |
| _____ 3. activation energy in a reaction | c. chemical change |
| _____ 4. rate of two reactions with different activation energies in a given medium | d. free energy |
| | e. reaction time |

Completion Write the correct name in the blank to complete each sentence below.

- During reactions, the breaking of a reactant is a process known as activation energy.
- The products of a reaction have less energy than the reactants.
- The process of the reaction is known as chemical change and making the matter a product is known as reaction.
- In a reaction, the energy that is released is known as exothermic reaction.

Part C: Integrating Drawings

Draw your response to each question in the space provided.

- Identify the chemical reaction in the figure.



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