Name:	Date:	
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Student Exploration: Limiting Reactants

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: chemical equation, chemical formula, chemical reaction, coefficient, limiting reactant, molecule, product, reactant, subscript

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

Imagine you and your friends are making hot dogs. A complete hot dog consists of a wiener and a bun. At the store, you buy four packages of eight wieners and three bags of 10 buns.

How many total hot dogs can you make?

30 hotdogs

Which ingredient limited the number of hot dogs you could make?

Buins.

Which ingredient will you have leftovers of?

Weiners.

Gizmo Warm-up

Just as ingredients can be put together to make a new food, substances can combine during a chemical reaction to produce new substances. The substances that undergo change are called reactants. The new substances are products.

Sometimes during a chemical reaction, one type of reactant will be used up before the other reactants. This reactant is the limiting reactant. Using the Limiting Reactants Gizmo, you can determine which reactant is limiting in various scenarios.

To begin, make sure H₂ + O₃ becomes H₂O is selected. The small "2" in H₃, O₃, and H₂O is a subscript. Subscripts represent the number of atoms in a molecule.

- 1. Use the sliders to set the number of O. molecules and H. molecules to two.
 - A. How many hydrogen molecules (H_s) are there?
 - B. How many oxygen molecules (O₂) are there?
- How many H₂O molecules do you think will form when these four molecules react?
- Click Play (). How many H₂O molecules actually formed?



- 2 molecules
- 2 molecules

Hydrogen atoms?

Oxygen atoms?

Gizmo Limiting Reactants Answer Key

Yi-Tong Ma

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