Activity C: Calculating specific heat	Get the Gizmo ready: • Click Reset.	
apcome near		

Introduction: The specific heat capacity of a substance is the amount of energy needed to change the temperature of that substance by 1 °C. Specific heat capacity can be calculated using the following equation: $q = mc\Delta T$

In the equation q represents the amount of heat energy gained or lost (in **joules**), m is the mass of the substance (in grams), c is the specific heat capacity of the substance (in J/g °C), and ΔT is the temperature change of the substance (in °C).

Goal: Calculate the specific heat capacities of copper, granite, lead, and ice.

- Solve: When you mix two substances, the heat gained by one substance is equal to the heat lost by the
 other substance. Suppose you place 125 g of aluminum in a calorimeter with 1,000 g of water. The water
 changes temperature by 2 °C and the aluminum changes temperature by -74.95 °C.
 - A. Water has a known specific heat capacity of 4.184 J/g °C. Use the specific heat equation to find out how much heat energy the water gained (q).

 $\triangle TQ = 1000 \times 4.18 \times 2Q = 8360\ J8360\ J; 8.4kJ$

B. Assume that the heat energy gained by the water is equal to the heat energy lost by the aluminum. Use the specific heat equation to solve for the specific heat of aluminum. (Hint: Because heat energy is lost, the value of q is negative.)

0.893

Aluminum's accepted specific heat value is 0.900 J/g °C. Use this value to check your work.

2. Calculate: Use the Gizmo to mix 200 g of copper at 100 °C with 1,000 g of water at 20 °C.

A. What is the final temperature? 2

21.42

B. Calculate the temperature change of each substance by subtracting the initial temperature from the final temperature.

 $\Delta T_{\rm matter?}$

1.49

 ΔT_{expect}

20

C. How much heat energy (q) did the water gain?

D. Now solve for the specific heat (c) of copper:

18

it is 20°C

Gizmo Calorimetry Lab Answers Key

H Kauffman

Gizmo Calorimetry Lab Answers Key:

This book delves into Gizmo Calorimetry Lab Answers Key. Gizmo Calorimetry Lab Answers Key is a vital topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and indepth insights into Gizmo Calorimetry Lab Answers Key, encompassing both the fundamentals and more intricate discussions.

- 1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Gizmo Calorimetry Lab Answers Key
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 - ∘ Chapter 5: Conclusion
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- 3. In chapter 2, the author will delve into the foundational concepts of Gizmo Calorimetry Lab Answers Key. The second chapter will elucidate the essential principles that must be understood to grasp Gizmo Calorimetry Lab Answers Key in its entirety.
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- 6. In chapter 5, this book will draw a conclusion about Gizmo Calorimetry Lab Answers Key. This chapter will summarize the key points that have been discussed throughout the book.
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