

Chemistry Form 5 Chapter 1

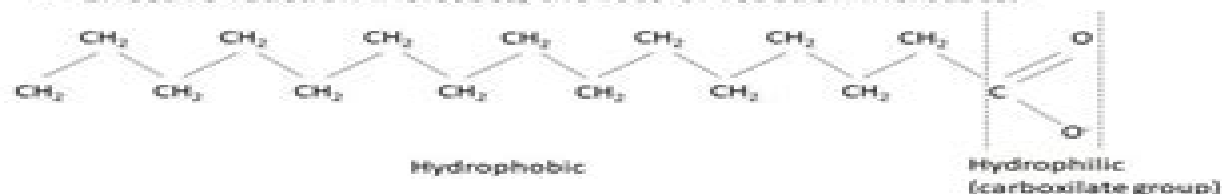
Rate of Reaction

- Rate of reaction is the change in quantity of reactant or product against time
- Method used to measure the volume of hydrogen gas is **water displacement method**

| <u>Fast reaction</u> | <u>Slow reaction</u> |
|-----------------------------------|----------------------------------|
| Short time taken for the reaction | Long time taken for the reaction |
| High rate of reaction | Low rate of reaction |
| Sodium + Water | Copper(II) oxide + Hydrogen gas |
| Neutralisation | Fermentation |
| Precipitation | Rusting |

Effective reaction

- Effective reaction are **collisions** between the **particles of reactants** that cause a reaction
- Occur when the particles of reactants have energy that is equal or higher than **activation energy**.
- Effective reaction increases, the rate of reaction increases.



Graph

Form Chemistry Notes

Candyce M. Jack, Patricia M. Watson

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the authors' names.

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Sixth Form Chemistry James W. McLaughlin, 1982

The Home Library of Useful Knowledge Richard S. Peale, 1886

Student Relationships Walter Gillian Clippinger, 1926

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