REVIEW QUESTIONS FOR FINAL EXAM (2nd Semester)

SECTION 12.1

Heat Transfer: $Q = mC\Delta T$ or $Q = mC(T_f - T_d)$

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and the same	2794	Miller	200
market and	2000	France.	4500
giorna .	100	territorio .	50000
inter .	2080	tables.	2000
Section 1	460		

1)Two solid objects are in contact as shown below.

Speed Speed

Heat will always flow from object A to object B if object B has a lower

a)mass d)temperature b)total energy

c)specific heat

e mone of the answers

For questions 2-3: The diagram below shows two solids with their temperatures (A=50°C, B=80°C).



2)What occurs when the two solids are placed in contact with each other as shown?

a)Heat energy flows from A to B. Solid A cools down.

b)Heat energy flows from A to B. Solid A gets hotter.
c)Heat energy flows from B to A. Solid B cools down.

d)Fleat energy flows from B to A. Solid B gets hotter.

3)The net flow of heat energy will stop from one object to the other when they both have the same

a)energy:

b)temperature

c)specific heat.

d)all the answers are correct

4))How much heat is transferred when a 0.2 kg piece of copper's temperature is raised from 50°C to 80°C?

Q=(0.2Kg)(385)/4+)(802-502)=(2310)

5)A 0.5 kg block of lead cools down from 30 °C to 20 °C. How much heat was transferred?

Q=(0.5%)(130J/m +)(20°C-30°C)

6)The heat transfer of a substance is 9400 J. The mass of this substance is 2 kg and its temperature rises from 80 °C to 100 °C. Based on the specific heat that you will calculate for this substance, what is it?

7)From the choices below, which material would be better for a pan handle?

d)trom 950

b)silver 235 e)zinc 388 e)copper 325 f)lead 130

8)Provide a brief description of absolute zero and their temperatures in Kelvin (K) and degrees Celsius (°C).

motion stops - Tc = -273°C, Tk = OK.

9)The boiling point of water is 100 °C. The boiling point of water in Kelvin (K) will be

TK = 273+Tc = 273+100 = 373K

10)A 1-liter of water taken out of a 5-gallon bucket full of water at 20 °C will

 a)register a lower temperature than that of the water left in the bucket

b)have less thermal energy than that of the water left in the bucket

chave tess specific heat than that of the water left in the bucket

$$T_c = T_K - 273 = 243K - 273 = -30°C$$

12)The kinetic molecular theory can explain the existence of thermal energy in a substance.

in)true

b)false

13)When using a thermometer for measuring the temperature of a substance, one should record the temperature

a)about 2 minutes after the thermometer has been placed in the substance. That is usually the time needed for the liquid inside to expand or shrink completely.

b)until thermal equilibrium has been reached

c)until all the heat from the substance-has been transferred to the thermometer

d)until the thermal energy from the substance is equal to the thermal energy of the thermometer

e all of the answers

14)A calorimeter contains 1 kg of water at 20 °C. A 0.20 kg block of zinc at 120 °C is placed in the water. What is the approximate final temperature of the system?

 $T_{F} = \frac{m_{A}c_{A}T_{Aj} + m_{B}c_{B}T_{Bi}}{m_{A}c_{A} + m_{B}c_{B}}$ $T_{F} = (1)(4180)(200) + (0.2)(388)(1200)$ (1)(4180) + (0.2)(388) $T_{F} = 83600 + 9312 = 21.8 °C$

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- 1. Understanding the eBook Final Physics Review Answers
 - The Rise of Digital Reading Final Physics Review Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Final Physics Review Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Final Physics Review Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Final Physics Review Answers
 - Personalized Recommendations
 - Final Physics Review Answers User Reviews and Ratings
 - Final Physics Review Answers and Bestseller Lists

- 5. Accessing Final Physics Review Answers Free and Paid eBooks
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 - Final Physics Review Answers eBook Subscription Services
 - Final Physics Review Answers Budget-Friendly Options
- 6. Navigating Final Physics Review Answers eBook Formats
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 - Final Physics Review Answers Compatibility with Devices
 - Final Physics Review Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
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 - Highlighting and Note-Taking Final Physics Review Answers
 - Interactive Elements Final Physics Review Answers
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 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Final Physics Review Answers
- 9. Balancing eBooks and Physical Books Final Physics Review Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Final Physics Review Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Final Physics Review Answers
 - Setting Reading Goals Final Physics Review Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Final Physics Review Answers
 - Fact-Checking eBook Content of Final Physics Review Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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