

D) the second column from the right.

25. An atom of oxygen has 8 protons, 9 neutrons, and 8 electrons. Its mass number is

- A) 8 B) 9 C) 16 D) 17

26. Substance X has 13 protons, 14 neutrons and 10 electrons. Determine the identity.

- A) ^{27}Mg B) ^{27}Ne C) $^{27}\text{Al}^{3+}$ D) ^{27}Al

Periodic Table

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
Q												T	U		V	W	X
	S																
R																Y	Z

27. Which of the following elements would form a 3^{+} cation?

- A) S B) T C) U D) V

28. Which of the following does not have a noble gas electron configuration?

- A) S^{2-} B) Na^{+} C) O^{2-} D) Al^{3+} E) P^{3-}

29. Which of the following is an ionic compound?

- A) HCl B) CH_4 C) CaS D) H_2O_2

30. When atoms form chemical bonds they can

- A) lose electrons B) gain electrons C) share electrons D) all of the above

31. A covalent bond would most likely form between a

- A) chlorine atom and a magnesium atom, B) chlorine atom and another chlorine atom,
C) sodium atom and a fluorine atom, D) chlorine atom and an iron atom.

32. Which group on the Periodic Table of the Elements contains elements that react with oxygen to form compounds with the general formula X_2O_7 ?

- A) Group 1 B) Group 14 C) Group 2 D) Group 16

33. Which of the following is true of ionic compounds?

- A) Oppositely charged ions attract each other. B) Electrons are shared between atoms.
C) Protons are transferred between atoms. D) Protons are shared between atoms.

34. Ba, Fe, and Cl are all examples of

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Jin-Ying Zhang



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Table of Contents Final Exam Gr10 Chemistry Scope

1. Understanding the eBook Final Exam Gr10 Chemistry Scope
 - The Rise of Digital Reading Final Exam Gr10 Chemistry Scope
 - Advantages of eBooks Over Traditional Books
2. Identifying Final Exam Gr10 Chemistry Scope
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Final Exam Gr10 Chemistry Scope
 - User-Friendly Interface
4. Exploring eBook Recommendations from Final Exam Gr10 Chemistry Scope
 - Personalized Recommendations
 - Final Exam Gr10 Chemistry Scope User Reviews and Ratings
 - Final Exam Gr10 Chemistry Scope and Bestseller Lists
5. Accessing Final Exam Gr10 Chemistry Scope Free and Paid eBooks
 - Final Exam Gr10 Chemistry Scope Public Domain eBooks
 - Final Exam Gr10 Chemistry Scope eBook Subscription Services

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 - Final Exam Gr10 Chemistry Scope Compatibility with Devices
 - Final Exam Gr10 Chemistry Scope Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Final Exam Gr10 Chemistry Scope
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 - Interactive Elements Final Exam Gr10 Chemistry Scope
- 8. Staying Engaged with Final Exam Gr10 Chemistry Scope
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Final Exam Gr10 Chemistry Scope
- 9. Balancing eBooks and Physical Books Final Exam Gr10 Chemistry Scope
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Final Exam Gr10 Chemistry Scope
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Final Exam Gr10 Chemistry Scope
 - Setting Reading Goals Final Exam Gr10 Chemistry Scope
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Final Exam Gr10 Chemistry Scope
 - Fact-Checking eBook Content of Final Exam Gr10 Chemistry Scope
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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