

# 17

## THERMOCHEMISTRY

### SECTION 17.1 THE FLOW OF ENERGY—HEAT AND WORK (pages 505–516)

This section explores the relationships between energy and heat, and discusses how between heat capacity and specific heat.

#### ► Energy Transformations (page 505)

1. What form of energy is associated in each case with the first transfer that occurs during chemical reactions? \_\_\_\_\_
2. Define the term *calorie*. Is calorie and (a) a potential energy, (b) a form of work done? \_\_\_\_\_
3. Circle the letter next to each sentence that is true about energy.
  - a. Energy is the capacity for doing work or supplying heat.
  - b. Energy is directional only because of its effects.
  - c. Heat is energy that transfers from one object to another because they are at the same temperature.
  - d. Gasoline contains a significant amount of chemical potential energy.
4. Circle the letter next to each sentence that is true about heat.
  - a. One effect of adding heat to a substance is an increase in the temperature of that substance.
  - b. Heat always flows from a cooler object to a warmer object.
  - c. If two objects remain in contact, heat will flow from the warmer object to the cooler object until the temperatures of both objects is the same.

#### ► Exothermic and Endothermic Processes (pages 506–507)

5. What can be considered the “system” and what are the “surroundings” when studying a solution of chemicals undergoing a reaction? Write your answers where indicated below.

System: \_\_\_\_\_

Surroundings: \_\_\_\_\_

# Fifty Frequently Forgotten Fun Facts Chemistry Answers

**E Durkheim**



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