

### D. Presence of a Catalyst

- What effect does the  $MnO_2$  catalyst have on the rate of evolution of  $O_2$  gas?
- Write a balanced equation for the decomposition of  $H_2O_2$ .

### E. Concentration of Reactants: Magnesium-Hydrochloric Acid System

Concentration of HCl	Volume of HCl	mol HCl	mass of Mg	mol Mg	mol HCl / mol Mg	Time (sec)
6 M	5 mL		0.0301			
4 M	5 mL		0.0188			
3 M	5 mL		0.0161			
1 M	5 mL		0.0124			

- Plot  $\frac{\text{mol HCl}}{\text{mol Mg}}$  (y-axis) versus time (x-axis). Instructor's approval of graph: \_\_\_\_\_
- From your graph, predict the time, in seconds, for 5 mg of Mg to react in 5 mL of 2.0 M HCl.

### F. Concentration of Reactants: Iodic Acid-Sulfurous Acid System

Molar concentration of  $HIO_3$  \_\_\_\_\_ Molar concentration of  $H_2SO_3$  \_\_\_\_\_

- What is the volume (mL) per drop of the  $HIO_3$  solution? \_\_\_\_\_ What is the volume (mL) per drop of the water? \_\_\_\_\_

Test Tube	Drops of $HIO_3$	mL $HIO_3$	Drops of $H_2O$	mL $H_2O$	mL $H_2SO_3$	$[HIO_3]_0$ (diluted)	Time (sec)
1					1.0		
2					1.0		
3					1.0		
4					1.0		
5					1.0		

- Plot  $[HIO_3]_0$  (y-axis) versus time (x-axis). Instructor's approval of graph: \_\_\_\_\_
- How does a change in the molar concentration of  $HIO_3$  affect the time required for the appearance of the deep-blue  $I_2$ -starch complex?
- Estimate the time, in seconds, for the deep-blue  $I_2$ -starch complex to form when 10 drops of 0.01 M  $HIO_3$  are used for the reaction. Assume all other conditions remain constant.

<sup>a</sup>See footnote 5.

# Factors Affecting Reaction Rate Lab Answers

**CO Houle**



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## **Table of Contents Factors Affecting Reaction Rate Lab Answers**

1. Understanding the eBook Factors Affecting Reaction Rate Lab Answers
  - The Rise of Digital Reading Factors Affecting Reaction Rate Lab Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Factors Affecting Reaction Rate Lab Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Factors Affecting Reaction Rate Lab Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Factors Affecting Reaction Rate Lab Answers
  - Personalized Recommendations
  - Factors Affecting Reaction Rate Lab Answers User Reviews and Ratings
  - Factors Affecting Reaction Rate Lab Answers and Bestseller Lists

5. Accessing Factors Affecting Reaction Rate Lab Answers Free and Paid eBooks
  - Factors Affecting Reaction Rate Lab Answers Public Domain eBooks
  - Factors Affecting Reaction Rate Lab Answers eBook Subscription Services
  - Factors Affecting Reaction Rate Lab Answers Budget-Friendly Options
6. Navigating Factors Affecting Reaction Rate Lab Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Factors Affecting Reaction Rate Lab Answers Compatibility with Devices
  - Factors Affecting Reaction Rate Lab Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Factors Affecting Reaction Rate Lab Answers
  - Highlighting and Note-Taking Factors Affecting Reaction Rate Lab Answers
  - Interactive Elements Factors Affecting Reaction Rate Lab Answers
8. Staying Engaged with Factors Affecting Reaction Rate Lab Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Factors Affecting Reaction Rate Lab Answers
9. Balancing eBooks and Physical Books Factors Affecting Reaction Rate Lab Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Factors Affecting Reaction Rate Lab Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Factors Affecting Reaction Rate Lab Answers
  - Setting Reading Goals Factors Affecting Reaction Rate Lab Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Factors Affecting Reaction Rate Lab Answers
  - Fact-Checking eBook Content of Factors Affecting Reaction Rate Lab Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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