

Question 4

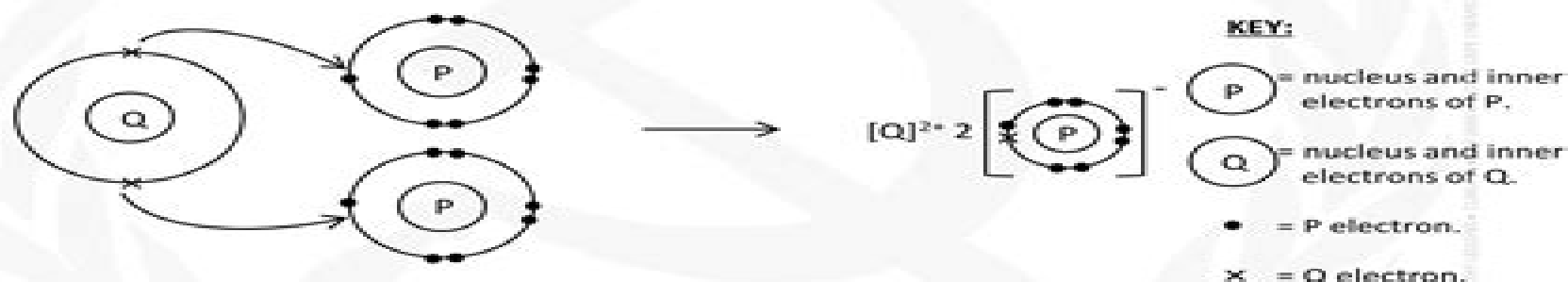
- 4(a) Radiocarbon dating using ^{14}C is a method of determining the age of once living, archaeological specimens which are no more than 50 000 years old.

^{127}I is an example of a substance which can be used as a tracer. Tracers which concentrate in some organs are introduced into the bodies of living organisms and used to study chemical changes.

- 4(b)(i) Number of occupied electron shells = period number.
Number of electrons in outer shell = group number.

| | P | Q | R | S |
|----------|---|---|---|---|
| Group # | 7 | 2 | 6 | 2 |
| Period # | 2 | 3 | 3 | 4 |

- 4(b)(ii) The compound is ionic.



Formula is QP_2 .

- 4(b)(iii) S will react more vigorously with hydrochloric acid than Q because S is lower in group 2 so it loses its outer electrons more easily than Q.

Equation for reaction of Q with HCl: $\text{Q}_{(s)} + 2\text{HCl}_{(aq)} \rightarrow \text{QCl}_{2(aq)} + \text{H}_{2(g)}$

F32chemistry June 2014 Paper

B Lingard



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