

100) Carbon-sulfur double bonds are not found in H_2S_2 molecules. What H_2S_2 molecule is a ring of eight atoms. How many atoms of sulfur are there in a molecule of H_2S_2 molecule?

$$\text{Ans} = \text{Total S atoms} = 2 \times 2 = 4 \quad \text{Total H atoms} = 2 \times 2 = 4$$

$$\text{H} = \text{H}_2\text{S}_2 = 4$$

Answer is 4 Ques 100

101) There only 2 nonmetals form covalent sulfur are not the Phosphorus.

102) Describe how can the Phosphorus form covalent.

Phosphorus has 5 valence electrons
It can form 5 covalent bonds


It can form 5 covalent bonds
Phosphorus Ques 102

103) Suggest other three can be other nonmetals form covalent bonds.

Carbon, Nitrogen, Oxygen
are other nonmetals
that can form covalent bonds Ques 103

104) Phosphorus molecule is a compound of sulfur (atomic number 16) and phosphorus.

Calculate the number of atoms in the molecule $\text{H}_2\text{S}_2\text{P}_2\text{O}_4$.

What is the molecular weight of $\text{H}_2\text{S}_2\text{P}_2\text{O}_4$?

Ans = 2 Ques 104



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