

Evidence for Chemical Change

One way of knowing that a chemical change has occurred is to observe that the properties of the products are different from those of the reactants. A new product can then become a reactant in another chemical reaction. In this experiment, you will observe a sequence of changes that occur when a solution that begins as copper(II) nitrate is subjected to a series of different reactions. All of the reactions will take place in the same test tube. At each step, you will look for evidence that a new substance has formed as a result of a chemical change. You will also observe energy changes and relate them to chemical reactions.

OBJECTIVES

Observe evidence that a chemical change has taken place.

Infer from observations that a new substance has been formed.

Identify and record observations that show energy is involved in chemical change.

Observe the color and solubility of some substances.

Describe reactions by writing word equations.

MATERIALS

- aluminum wire, 12 cm
- beaker, 100 mL
- Bunsen burner and related equipment
- copper(II) nitrate, 0.75 M
- glass stirring rod
- gloves
- HCl, 3.0 M
- iron ring
- lab apron
- lab marker
- NaOH, 3.0 M
- ring stand
- ruler
- safety goggles
- test tube, 13 mm × 100 mm
- test-tube rack
- wire gauze, ceramic-centered



Always wear safety goggles and a lab apron to protect your eyes and clothing. If you get a chemical in your eyes, immediately flush the chemical out at the eyewash station while calling to your teacher. Know the location of the emergency lab shower and the eyewash station and the procedures for using them.



Do not touch any chemicals. If you get a chemical on your skin or clothing, wash the chemical off at the sink while calling to your teacher. Make sure you carefully read the labels and follow the precautions on all containers of chemicals that you use. If there are no precautions stated on the label, ask your teacher what precautions to follow. Do not taste any chemicals or items used in the laboratory. Never return leftover chemicals to their original containers; take only small amounts to avoid wasting supplies.

Evidence For Chemical Change Lab Answers A5

R Sandford



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Chemfile Skills Practice Experiments Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2006 **Chemistry in the Laboratory** Jo A. Beran, 1996 This updated and revised chemistry manual provides a well rounded understanding of concepts in the general chemistry laboratory Utilising visual aids experiments and equipment are explained and results and their pertinence discussed Selected Water Resources Abstracts ,1974 **Scientific and Technical Aerospace Reports** ,1988 Nuclear Science Abstracts ,1975-12 Physical and Chemical Hydrogeology Patrick A. Domenico, Franklin W. Schwartz, 1997-10-15 This book shows readers how to apply hydrogeology principles to a host of problems related to water supply contamination and energy resources It discusses hydraulic testing modeling of contaminant transport process and parameter determination and remediation It also addresses porosity permeability and flow for continental environments marine environments and the borders between them *Cumulated Index Medicus* ,1972

Polymers Bryan Ellis, Ray Smith, 2008-10-29 A reliable source for scientific and commercial information on over 1 000 polymers this revised and updated edition features 25 percent new material including 50 entirely new entries that reflect advances in such areas as conducting polymers hydrogels nano polymers and biomaterials The second edition also comes with unlimited access to a complete fully searchable web version of the reference Powerful retrieval software allows users to customize their searches and refine results Each entry includes trade names properties manufacturing processes commercial applications supplier details references and links to constituent monomers Index Medicus ,2004 Vols for 1963 include as pt 2 of the Jan issue Medical subject headings **The Laboratory Digest** ,1960 **Government Reports Announcements & Index** ,1988 **The Journal of the Iron and Steel Institute** Iron and Steel Institute, 1958 Includes sect A survey of literature on the manufacture and properties of iron and steel and kindred subjects title varies **Journal** ,1958 Includes the institute s Proceedings *Abstracts* Materials Research Society. Fall Meeting, 1992 The History of Neuroscience in Autobiography Larry R. Squire, 1996 The History of Neuroscience in Autobiography is a book series edited by Tom Albright and Larry R Squire that details the lives and discoveries of eminent senior neuroscientists During his term as president of the Society for Neuroscience in 1993 94 Dr Squire initiated the collection of these autobiographies from leading neuroscientists so they could share their personal narratives These delightful often humorous and touching narratives allow each scientist to discuss their lives and the forces that shaped the path to their prominent careers and discoveries publisher s website Chemical Abstracts ,1991 Federation Proceedings Federation of American Societies for Experimental Biology, 1960 Vols for 1942 include proceedings of the American Physiological Society **Papers Presented at the ... Meeting** American Chemical Society. Division of Polymer Chemistry, 1991 **Current Advances in Protein Biochemistry** ,1994 *Toxicology Abstracts* ,1991 Includes annual author and subject indexes

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