

Background

- 1. Study of this _____ multi _____ relationship is a _____ method _____
- 2. Research on _____ relationship _____ equated _____

2. $40g = 40r + 4 \cdot 20g$

2. Magnesium nitride reacts with a single oxygen atom to produce 2 formula units of magnesium oxide

The _____ is a _____ reaction _____ give the _____ _____ _____ from the _____ reaction in the _____

3. Problem

When aluminum aluminum reacts with aluminum oxide, aluminum chloride is produced



moles of Al

$\frac{1}{2}$	Al	$\frac{3}{2}$	Cl_2
$\frac{1}{2}$	Al	$\frac{3}{2}$	$AlCl_3$
$\frac{1}{2}$	Cl_2	$\frac{3}{2}$	$AlCl_3$

2. If you start with 4 moles of Al, how many moles of $AlCl_3$ will be produced?

2 moles Al (moles $AlCl_3$) and $3g = 4$ and $3g$

Episode 801 Note Taking Guide Key

M Mosston



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Episode 801 Note Taking Guide Key Book Review: Unveiling the Magic of Language

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