

Calculate the enthalpy of combustion of ethane, C_2H_6 , at 25°C and one atmosphere pressure.



Compound	$\Delta H_f^\circ, \text{kJ}\cdot\text{mol}^{-1}$
$\text{C}_2\text{H}_6(g)$	-84.7
$\text{H}_2\text{O}(l)$	-285.8
$\text{CO}_2(g)$	-393.5

Entropy For Combustion Of Ethane C₂H₆

Michael Rajnik



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