

ENGINEERING MECHANICS
STATICS

J. L. MERIAM
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9th Edition

Solution Manual



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Mechanical Engineering Statics Second Edition Solution Manual

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 $2\text{ KClO}_3 \rightarrow 2\text{ KCl} + 3\text{ O}_2$
 $2\text{ NaCl} + \text{F}_2 \rightarrow 2\text{ NaF} + \text{Cl}_2$
 $2\text{ H}_2 + \text{O}_2 \rightarrow 2\text{ H}_2\text{O}$
 $5\text{ Pb} + 2\text{ HCl} \rightarrow 2\text{ H}_2 + \text{PbCl}_2$
 $2\text{ Al} + 3\text{ Br}_2 \rightarrow 2\text{ AlBr}_3$
 $3\text{ K}_2\text{SO}_4 + 6\text{ KBr} \rightarrow 1\text{ Al}_2\text{SO}_4 + 3\text{ Br}_2$
 $4\text{ CH}_4 + 2\text{ O}_2 \rightarrow 2\text{ CO}_2 + 2\text{ H}_2\text{O}$
 $8\text{ C} + 3\text{ H}_2 + 5\text{ O}_2 \rightarrow 2\text{ CO}_2 + 4\text{ H}_2\text{O}$

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 $1\text{ N}_2 + \text{F}_2 \rightarrow 2\text{ NF}_3$
 $2\text{ C}_6\text{H}_{10} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
 $3\text{ HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$
 $\text{GaBr}_3 + \text{NaBr} \rightarrow \text{SNO}_4 + \text{NaSO}_3 + \text{Ga}_2\text{SO}_3 + 3\text{ NF}_3 + \text{SnF}_2 + \text{N}_2\text{O}_3$ using the equation from problem 2 above answer the following

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 $1\text{ N}_2 + 3\text{ F}_2 \rightarrow 2\text{ NF}_3$
 $2\text{ C}_6\text{H}_{10} + 17\text{ O}_2 \rightarrow 12\text{ CO}_2 + 10\text{ H}_2\text{O}$
 $3\text{ HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$
 $2\text{ GaBr}_3 + 3\text{ Na}_2\text{SO}_3 \rightarrow \text{Ga}_2\text{SO}_3 + 3\text{ NaBr} + 3\text{ SNO}_4 + 2\text{ NF}_3 + 3\text{ SnF}_2 + 1\text{ N}_2\text{O}_3$ using the following equation
 $2\text{ NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{ H}_2\text{O} + \text{Na}_2\text{SO}_4$

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 $\text{Ce} + \text{Zn} + \text{CuCl}_2 \rightarrow \text{ZnCl}_2 + \text{Cu}$
 how many moles of Ce ZnCl_2 ZnCl_2 will be produced from 23.0 g of Ce 23.0 g of Zn

assuming $\text{CeCl}_2 \cdot 2\text{H}_2\text{O}$ is available in excess

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 $\text{KBr} + \text{CO}_2 \rightarrow \text{K}_2\text{CO}_3 + \text{C}$
 $\text{GaBr}_3 + \text{Na}_2\text{SO}_3 \rightarrow \text{Ga}_2\text{O}_3 + \text{SO}_2 + \text{NaBr}$
using the following equation
 $\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow \text{Na}_2\text{SO}_4 + \text{H}_2\text{O}$

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 $\text{GaBr}_3 + \text{Na}_2\text{SO}_3 \rightarrow \text{Ga}_2\text{O}_3 + \text{SO}_2 + \text{NaBr}$
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same thing goes with the o3 i am so lost

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web it consists of the following steps describe the equipment details take pre maintenance safety precautions check the blower area check the filter area check the cooling area check the condenser compressor area check

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