

21. In which of the following reactions addition does not take place according to Markovnikov's rule ?
(IIT 2000)
- (a) $\text{CH}_3\text{CH}=\text{CH}_2 + \text{HCl} \xrightarrow{\text{ROOR}}$
(b) $\text{CH}_3\text{CH}=\text{CH}_2 + \text{HBr} \xrightarrow{\text{ROOR}}$
(c) $\text{CH}_2=\text{CH}_2 + \text{HI} \xrightarrow{\text{ROOR}}$
(d) $\text{CH}_3\text{CH}=\text{CHCH}_3 + \text{HBr} \xrightarrow{\text{ROOR}}$
22. Ozonolysis of 2-methyl butene-2 yields : (UPC 2003)
- (a) ketone (b) aldehyde
(c) aldehyde & ketone (d) none
23. C_6H_{12} on ozonolysis gave only propanone. Alkene is :
(AFMC 2003)
- (a) $\text{CH}_3\text{CH}_2\text{CH}=\text{C}(\text{CH}_3)_2$
(b) $(\text{CH}_3)_2\text{C}=\text{CH}-\text{CH}_3$
(c) $(\text{CH}_3)_2\text{C}=\text{C}(\text{CH}_3)_2$
(d) $\text{CH}_3\text{CH}_2\text{CH}=\text{CHCH}_2\text{CH}_3$
24. The reaction,
 $\text{CH}_2=\text{CHCH}_3 + \text{HBr} \longrightarrow \text{CH}_3\text{CHBrCH}_3$ is :
(AIIMS 2003)
- (a) Nucleophilic addition
(b) Electrophilic addition
(c) Electrophilic substitution
(d) Free radical addition
25. Which of the following alkenes is most stable ?
(UPC 2003)
- (a) $\text{R}_2\text{C}=\text{CR}_2$ (b) $\text{R}-\text{CH}=\text{CH}-\text{R}$
(c) $\text{RCH}=\text{CH}_2$ (d) $\text{CH}_2=\text{CH}_2$
26. A molecule of HCl adds on 3-methyl but-1-ene. Product formed is : (IIT 2001)
- (a) $\text{CH}_3\text{CH}(\text{CH}_3)-\text{CHCl}-\text{CH}_3$
(b) $\text{CH}_3-\text{C}(\text{Cl})(\text{CH}_3)-\text{CH}_2-\text{CH}_3$
(c) $\text{CH}_3\text{CH}(\text{CH}_3)-\text{CH}_2-\text{CH}_2\text{Cl}$
(d) $\text{CH}_3\text{CH}(\text{CH}_3)-\text{CH}(\text{Cl})-\text{CH}_2\text{CH}_3$
27. The intermediate during the addition of HCl to propene in presence of peroxide is : (IIT 2002)
- (a) $\text{CH}_3\dot{\text{C}}\text{HCH}_2\text{Cl}$ (b) $\text{CH}_3\dot{\text{C}}\text{HCH}_3$
(c) $\text{CH}_3\text{CH}_2\dot{\text{C}}\text{H}_2$ (d) $\text{CH}_3\text{CH}_2\dot{\text{C}}\text{H}$
28. Which of the following exhibit *cis, trans*-isomerism ?
(a) $\text{CH}_2\text{Br}-\text{CH}_2\text{Br}$
- (b) $\text{CBr}_2=\text{CH}_2$
(c) $\text{CHBr}=\text{CHBr}$
(d) $\text{CBr}_2=\text{CH}-\text{CH}_3$
29. The compound that dehydrates most readily is :
(CPMT 2001)
- (a) R_2CHOH (b) RCH_2OH
(c) R_3COH (d) $\text{R}-\text{OH}$
30. The addition of propene with HOCl proceeds via the addition of : (IIT 2001)
- (a) H^+ in the first step
(b) Cl^+ in the first step
(c) OH^- in the first step
(d) Cl^+ and OH^- in a single step
31. Which compound is sure to be alkene ?
(AFMC 2003)
- (a) C_2H_4 (b) C_3H_6
(c) C_4H_8 (d) C_5H_{10}
(e) All
32. 1-chlorobutane with alc. KOH gives : (AFMC 2001)
- (a) 1-butene (b) 1-butanol
(c) 2-butene (d) 2-butanol
33. How much ethylene is oxidised by 3 litres of O_2 under identical conditions ? [CBSE (Med.) 2002]
- (a) One mole (b) One litre
(c) 3 litre (d) 0.3 litre
34. Which one of the following alkenes will react fastest with H_2 under catalytic hydrogenation condition ?
(IIT (S) 2000)
- (a) $\text{R}_2\text{C}=\text{CH}_2$ (b) $\text{RHC}=\text{CHR}$
(c) $\text{R}_2\text{C}=\text{CHR}$ (d) $\text{R}_2\text{C}=\text{CR}_2$
35. Which of the following gases gives a red precipitate with ammonical solution of cuprous chloride ?
(DCE 2004)
- (a) C_2H_4 (b) C_2H_6
(c) C_2H_2 (d) $\text{CH}_3\text{CH}_2\text{CH}_3$
36. Which will not give a precipitate with Tollen's reagent?
(DCE 2002)
- (a) Ethyne
(b) 2-pentyne
(c) 1-butyne
(d) 3-methyl-1-butyne
37. Which hydrocarbon reacts with sodium in liquid NH_3 ?
(AIEEE 2008)
- (a) $\text{CH}_3-\text{CH}_2-\text{C}\equiv\text{CH}$
(b) $\text{CH}_3-\text{C}\equiv\text{C}-\text{CH}_3$
(c) $\text{CH}_3-\text{CH}=\text{CH}-\text{CH}_3$
(d) $\text{CH}_3-\text{CH}_2-\text{CH}=\text{CH}_2$
38. Increasing order of volatility of C_2H_6 , C_2H_4 , C_2H_2 and C_6H_6 is :
(BHU 2003)

Mcqs Of Hydrocarbons

RJ Shavelson



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of Green Chemistry the latest trend for sustainable chemistry as Appendix II

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Table of Contents Mcqs Of Hydrocarbons

1. Understanding the eBook Mcqs Of Hydrocarbons
 - The Rise of Digital Reading Mcqs Of Hydrocarbons
 - Advantages of eBooks Over Traditional Books
2. Identifying Mcqs Of Hydrocarbons
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mcqs Of Hydrocarbons
 - User-Friendly Interface
4. Exploring eBook Recommendations from Mcqs Of Hydrocarbons
 - Personalized Recommendations
 - Mcqs Of Hydrocarbons User Reviews and Ratings
 - Mcqs Of Hydrocarbons and Bestseller Lists
5. Accessing Mcqs Of Hydrocarbons Free and Paid eBooks
 - Mcqs Of Hydrocarbons Public Domain eBooks
 - Mcqs Of Hydrocarbons eBook Subscription Services
 - Mcqs Of Hydrocarbons Budget-Friendly Options
6. Navigating Mcqs Of Hydrocarbons eBook Formats
 - ePub, PDF, MOBI, and More
 - Mcqs Of Hydrocarbons Compatibility with Devices
 - Mcqs Of Hydrocarbons Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mcqs Of Hydrocarbons
 - Highlighting and Note-Taking Mcqs Of Hydrocarbons
 - Interactive Elements Mcqs Of Hydrocarbons
8. Staying Engaged with Mcqs Of Hydrocarbons

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mcqs Of Hydrocarbons
9. Balancing eBooks and Physical Books Mcqs Of Hydrocarbons
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mcqs Of Hydrocarbons
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Mcqs Of Hydrocarbons
- Setting Reading Goals Mcqs Of Hydrocarbons
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Mcqs Of Hydrocarbons
- Fact-Checking eBook Content of Mcqs Of Hydrocarbons
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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