

Chemistry 2 ACS final exam 2024 Comprehensive Review

ideal gas law conditions - **Answer>>** low pressure, high temperature

boyle's law - **Answer>>** $p_1v_1=p_2v_2$

gases are highly _____ - **Answer>>** compressible

gases expand _____ to fill the container they are placed in - **Answer>>** spontaneously

the distance between molecules in gases are _____ - **Answer>>** far apart in gases

1 mol of any gas at STP is - **Answer>>** 22.4 liters

the lighter the gas, _____ - **Answer>>** the faster the effusion rate

the deviation from ideal behavior of a gas is most evident at high _____ and low _____ - **Answer>>** pressure, temp

Non reacting gas mixtures are homongeneous - **Answer>>**

normal melting point at - **Answer>>** 760 torr or 1 atm

the kinetic-molecular theory predicts that pressure rises as the temperature of a gas increases because _____ -

Answer>> both the gas molecules collide more frequently with the wall and the gas molecules collide more energetically with the wall

Final Exam Review Packet 2 Chemistry

Pamela Primrose



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