

Exam I Study Guide

- **Physical Changes vs. Chemical Changes (Can you undo it?)**
 - Physical changes result in changes of appearance, but not composition.
 - Melting, evaporation, freezing, etc.
 - Chemical changes result in changes of composition of matter. (Chemical reaction)
 - Burning, mixing, dissolving, etc.
- **Physical Properties vs. Chemical Properties**
 - Physical properties can be described by observing the matter without a change in the composition.
 - Color, odor, density, temperature, etc.
 - Chemical properties can only be detected by altering the composition.
 - Flammability, oxidization, etc.
- **Atoms vs. Molecules**
 - Atoms are the smallest unit of a given element that carry the properties of that element.
 - Molecules are composed of two or more atoms.
- **Elements vs. Compounds**
 - Elements occur in nature and cannot be broken down to simpler particles through ordinary means.
 - Compounds are the combination of two or more elements.
- **Accuracy vs. Precision**
 - Accuracy refers to the agreement of a measurement with a known or accepted value. (Bullseye)
 - Correctness
 - Precision refers to the agreement of a measurement with other measurements in a given set of measurements. (Missing the bullseye but landing all darts in one place)
 - Repeatability or reproducibility
- **Rules Concerning Significant Figures**
 - All nonzero digits ARE significant.
 - Zeros between nonzero digits ARE significant.
 - Zeros at the beginning of a number are NOT significant; they only indicate the position of the decimal point.
 - Zeros at the end of a number to the right of a decimal point ARE significant.
 - Zeros at the end of a number with no decimal point are ASSUMED to be NOT significant, UNLESS we indicate their significance by placing a line over the last zero that is significant.

General Chemistry Final Study Guide

Phyllis Blumberg



General Chemistry Final Study Guide:

Study Guide for Ebbing/Gammons General Chemistry Darrell Ebbing, Steven D. Gammon, 2016-02-05 The perfect way to prepare for exams build problem solving skills and get the grade you want This guide reinforces your understanding of major concepts learning objectives and key terms presented in your text and further develops your problem solving skills Each chapter features both a diagnostic pre test and post test additional practice problems and their worked out solutions and cumulative unit exams

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