

Chapter 25 - Nuclear Chemistry

The study of _____ and their uses in chemistry

Radiation -

- The penetrating rays & particles emitted by a radioactive source

Radioactivity -

- Process that occurs when nuclei change _____

Radioisotopes -

- Atoms containing _____
➤ Recall... What are isotopes?

○

How do nuclear reactions differ from chemical reactions?

Chemical Reactions

- Atoms become stable by _____ or _____ electrons
- Rate _____ changed by catalysts, temperature, pressure, etc.
- Mass & charge _____

Nuclear Reactions

- Nuclei become stable by undergoing changes that _____ of energy
- Rate _____ be changed
- Charge conserved, mass _____
○ A _____ small amount of mass is converted into energy

Nuclear Stability

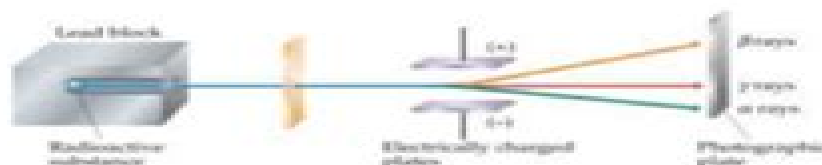
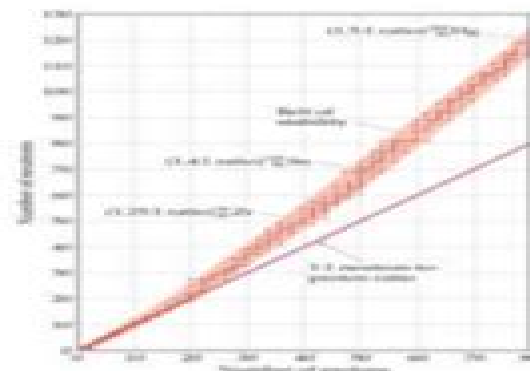
- Most nuclei are stable
- The stability of a nucleus depends on its neutron to proton _____
- The stable nuclei are in the ' _____ '

Radioactive Decay

- Occurs in atoms with too _____ or too _____ neutrons
- The nucleus will attempt to become more stable by releasing energy through _____
- Transmutation -
○ when the unstable nucleus of one element is _____ into a stable nucleus of a *different* element
○ Occurs naturally and artificially

Types of Radioactive Decay

- Alpha particles,
- Beta particles,
- Positron emission
- Gamma rays,



General Chemistry Chapter 25 Nuclear Chemistry

Lei Shi



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texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields **Handbook of Nuclear Chemistry** Attila Vértes,Sándor Nagy,Zoltán Klencsár,Rezso György Lovas,Frank Rösch,2010-12-10 This revised and extended 6 volume handbook set is the most comprehensive and voluminous reference work of its kind in the field of nuclear chemistry The Handbook set covers all of the chemical aspects of nuclear science starting from the physical basics and including such diverse areas as the chemistry of transactinides and exotic atoms as well as radioactive waste management and radiopharmaceutical chemistry relevant to nuclear medicine The nuclear methods of the investigation of chemical structure also receive ample space and attention The international team of authors consists of scores of world renowned experts nuclear chemists radiopharmaceutical chemists and physicists from Europe USA and Asia The Handbook set is an invaluable reference for nuclear scientists biologists chemists physicists physicians practicing nuclear medicine graduate students and teachers virtually all who are involved in the chemical and radiopharmaceutical aspects of nuclear science The Handbook set also provides further reading via the rich selection of references *General Chemistry* Pierce Wilson Selwood,1954

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