

Question	Answer	Marks	Guidance
5 (a) (i)	<p><i>Nuclear charge mark</i> (Across the period) number of protons increases OR greater nuclear charge ✓</p> <p><i>Quality of written communication – nuclear OR proton(s) OR nucleus spelled correctly ONCE for the first marking point</i></p> <p><i>Distance / shielding mark</i> (Outermost) electrons are in the same shell OR (Outermost) electrons experience the same shielding OR Atomic radius decreases ✓</p> <p><i>Nuclear attraction (to electron) mark</i> Greater nuclear attraction (on outermost electrons) OR (outer) electrons are attracted more strongly (to the nucleus) ✓</p>	3	<p>FULL ANNOTATIONS WITH TICKS, CROSSES, CON, etc MUST BE USED</p> <p>Comparison should be used for each mark</p> <p>IGNORE atomic number increases, but ALLOW proton number increases IGNORE nucleus gets bigger IGNORE 'effective nuclear charge increases' DO NOT ALLOW 'charge' increases without reference to nuclear</p> <p>ALLOW shielding is similar BUT IGNORE 'there is shielding' DO NOT ALLOW sub-shells OR orbitals</p> <p>ALLOW greater nuclear pull for greater nuclear attraction DO NOT ALLOW use of greater nuclear charge for greater nuclear attraction for third mark</p>
	(ii) (Diamond and graphite form) gaseous atoms (of carbon when they are ionised) ✓	1	ALLOW the atoms are in the gaseous state

F321 Jun 2013 Paper

Peter Cann, Peter Hughes



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Cambridge International AS and A Level Chemistry Peter Cann, Peter Hughes, 2015-03-06 Endorsed by Cambridge Assessment International Education for full syllabus coverage Foster a deeper understanding of theoretical concepts through clear guidance and opportunities for self assessment throughout covers the entire Cambridge International AS A Level Chemistry syllabus 9701 Navigate the different routes through the course with ease with clearly divided sections for AS and A Level Focus learning with learning outcomes clearly defined at the beginning of each section Test knowledge and understanding with past paper and exam style questions Address the Key Concepts in the syllabus which are clearly highlighted throughout the course The Revision and Practice CD included with every Student s Book provides interactive tests summaries of each topic and advice on examination techniques

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Table of Contents F321 Jun 2013 Paper

1. Understanding the eBook F321 Jun 2013 Paper
 - The Rise of Digital Reading F321 Jun 2013 Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying F321 Jun 2013 Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an F321 Jun 2013 Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from F321 Jun 2013 Paper
 - Personalized Recommendations
 - F321 Jun 2013 Paper User Reviews and Ratings
 - F321 Jun 2013 Paper and Bestseller Lists
5. Accessing F321 Jun 2013 Paper Free and Paid eBooks
 - F321 Jun 2013 Paper Public Domain eBooks
 - F321 Jun 2013 Paper eBook Subscription Services
 - F321 Jun 2013 Paper Budget-Friendly Options
6. Navigating F321 Jun 2013 Paper eBook Formats

- ePub, PDF, MOBI, and More
 - F321 Jun 2013 Paper Compatibility with Devices
 - F321 Jun 2013 Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of F321 Jun 2013 Paper
 - Highlighting and Note-Taking F321 Jun 2013 Paper
 - Interactive Elements F321 Jun 2013 Paper
8. Staying Engaged with F321 Jun 2013 Paper
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers F321 Jun 2013 Paper
9. Balancing eBooks and Physical Books F321 Jun 2013 Paper
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection F321 Jun 2013 Paper
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine F321 Jun 2013 Paper
- Setting Reading Goals F321 Jun 2013 Paper
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of F321 Jun 2013 Paper
- Fact-Checking eBook Content of F321 Jun 2013 Paper
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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