

Covalent compounds- consist of nonmetal only.

If chemical formula starts with H- it's an acid: find its name or formula in the Acids List.
Exception: H_2O and H_2O_2 .

If chemical formula consist of 2 nonmetals only- use the following naming rules.

Ex: name the following compound - H_2O_2

(a) name the first element- unchanged name from PT

- hydrogen

(b) for the second element, take its root and add ending -ide

- oxide

(c) add prefixes which will show how many atoms of each element are here in a compound

Answer: **dihydrogen dioxide**

Write the names for each of the following covalent compounds-

(1) NO - nitrogen monoxide

(2) HNO_3 - nitric acid

(3) H_3PO_4 _____

(4) N_2O_5 _____

(5) SiO_2 _____

(6) N_2O_5 _____

(7) HNO_2 _____

(8) HBr _____

(9) C_2H_6 _____

(10) H_2SO_3 _____

(11) CCl_4 _____

(12) $\text{HC}_2\text{H}_3\text{O}_2$ _____

(13) HClO_4 _____

(14) HClO_3 _____

(15) H_2SO_4 _____

(16) H_2S _____

(17) HI _____

(18) H_2O_2 _____

(19) Cl_3F _____

(20) N_2H_4 _____

(21) Cl_2O_7 _____

(22) HCl _____

(23) NH_3 _____

(24) H_2SiO_4 _____

(25) Cl_2O_3 _____

(26) PCl_3 _____

(27) HF _____

(28) Cl_2O _____

(29) CS_2 _____

(30) H_2CO_3 _____

(31) SF_6 _____

(32) SiC _____

(33) P_2S_5 _____

(34) F_2O_2 _____

Naming Ionic Compounds Composition Answer Key

Bill W. Tillery



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