

Mass Relationships in Chemical Equations

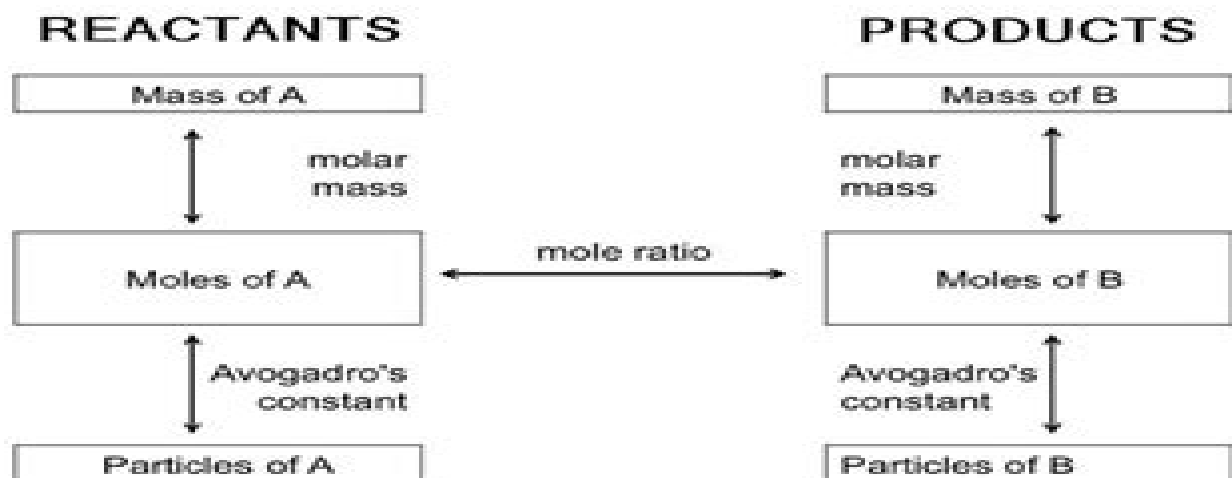
Mole Ratio: indicates the relationship between the number of moles of different chemicals used or produced in chemical reactions. (This information comes from the coefficients of a balanced chemical equation).



mole ratios: $\frac{1 \text{ mol N}_2}{3 \text{ mol H}_2}$ or $\frac{3 \text{ mol H}_2}{2 \text{ mol NH}_3}$

Knowledge of chemical equations and moles can be used to calculate mass relationships among the substances involved in a chemical reaction

Strategies for Solving Mass Relationships in Chemical Equations



Mass And Mole Relationships In A Chemical Reaction

Lab Answers

Joacim Rocklöv



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