

### Ionic Naming Worksheet #2

Name the following ionic compounds:

- 1)  $\text{Mo}(\text{SO}_4)_3$  \_\_\_\_\_
- 2)  $\text{Ti}(\text{NO}_3)_3$  \_\_\_\_\_
- 3)  $\text{NaH}$  \_\_\_\_\_
- 4)  $\text{BeSO}_3 \cdot 7\text{H}_2\text{O}$  \_\_\_\_\_
- 5)  $\text{Fe}_3(\text{PO}_4)_2$  \_\_\_\_\_
- 6)  $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$  \_\_\_\_\_
- 7)  $\text{V}(\text{H}_2\text{PO}_4)_4$  \_\_\_\_\_
- 8)  $\text{Mn}_3(\text{PO}_4)_7$  \_\_\_\_\_
- 9)  $\text{Fe}(\text{CN})_3$  \_\_\_\_\_
- 10)  $\text{BaSO}_4$  \_\_\_\_\_

Write the formulas for the following ionic compounds:

- 11) vanadium (V) arsenide \_\_\_\_\_
- 12) cobalt (III) nitrite \_\_\_\_\_
- 13) yttrium perchlorate \_\_\_\_\_
- 14) silver phosphide \_\_\_\_\_
- 15) tin (IV) selenide \_\_\_\_\_
- 16) cadmium sulfite \_\_\_\_\_
- 17) nickel (III) sulfate \_\_\_\_\_
- 18) scandium hydroxide \_\_\_\_\_
- 19) copper (I) oxide \_\_\_\_\_
- 20) ammonium nitride \_\_\_\_\_

# Ionic Bonding Worksheet Instructional Fair

**Margaret Irene Lindsay**



### **Ionic Bonding Worksheet Instructional Fair:**

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