

Covalent Bonding and Ionic Bonding Exam Questions with Correct Answers

Covalent bonds are bonds between 2 or more....? - Answer-Nonmetals.

What do covalent bonds form? - Answer-Molecules.

True or False. Covalent bonds form compounds. - Answer-False, they only form molecules.

What is shared in a covalent bond? - Answer-The valence electrons.

What happens when the valences electrons are shared in a covalent bond? - Answer-Sometimes the sharing is equal and non-polar.

What are the 7 diatomic molecules? - Answer-O₂, F₂, N₂, Cl₂, H₂, Br₂, I₂.

What happens when nonmetals are not able to gain electrons to complete their octet? - Answer-They don't remain unstable, but they share valence electrons in a covalent bond.

What is a covalent bond? - Answer-The attraction that forms when atoms share valence electrons.

What is it called in a covalent bond when the sharing is unequal? - Answer-Polar.

What does polar covalent bonds make? - Answer-A positive and a negative end on the molecule.

What is the rule that says atoms want to have 8 valence electrons to be stable? - Answer-The Octet Rule

What is Ionization? - Answer-When atoms gain or lose electrons to be stable.

Ionic Covalent Bonding Test Study Guide Answers

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