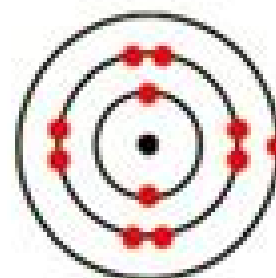
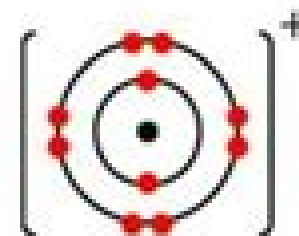


# Ionic bonding

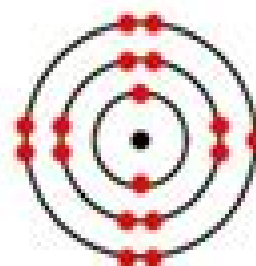
- In ionic bonding atoms transfer electrons, forming ions.
- Positively and negatively charged ions attract each other in giant ionic lattices.



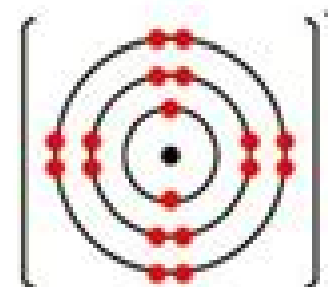
sodium atom,  
Na 2,8,1



sodium ion  
Na [2,8]<sup>+</sup>



chlorine atom,  
Cl 2,8,7



chloride ion,  
Cl<sup>-</sup> [2,8,8]<sup>-</sup>

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**Rachel Turoscy**



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