

Reminders:

1. In a neutral atom the number of protons equals the number of electrons.
2. An atom can NEVER gain or lose protons.
3. The number of protons equals the atomic number

Ion Practice Set

1. What is an ion?

Es un átomo o un grupo de átomos (molécula) con una carga negativa o positiva

2. What does the number next to the ions signify?

La carga que llevan

Complete the following table, using the periodic table in the back of your book.

	ELEMENT NAME	ION SYMBOL	NUMBER OF PROTONS	NUMBER OF ELECTRONS	NUMBER OF ELECTRONS LOST OR GAINED
ex	Fluorine	F ⁻	9	10	gained one
1	Yodo	I ⁻	53	54	Gained one
2	Azufre	S ⁻²	16	18	gained two
3	potassium	K ⁺	19	18	lost one
4	Calcio	Ca ⁺²	20	18	Lost two
5	Bromo	Br ⁻	35	36	Gained one
6	Estroncio	Sr ⁺²	38	36	Lost two
7	Hidrogeno	H ⁺	1	2	Gained one
8	Oxígeno	O ⁻²	8	10	gained two
9	Magnesio	Mg ⁺²	12	10	lost two
10	aluminum	Al ⁺²	13	10	Lost two
11	Selenio	Se ⁻²	34	36	Gained two
12	Hidrogeno	H ⁺	1	0	Lost one
13	lithium	Li ⁺	3	2	lost one
14	Rubidio	Rb ⁻	37	38	Gained one
15	Cloro	Cl ⁻	17	18	Gained one

Ion Practice Set Answer Key

Richard Bailey

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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