

FINAL EXAM REVIEW

Student Name: _____

Period: _____

PERSONAL FINANCE

Course #6093

Mrs. Lee Hartman

DATE OF FINAL EXAM TEST: _____

1. Please refer to your copy of each of the Chapter Reviews at the end of each chapter in your class workbook...
2. Any Make-Up work must be turned in by 3:30 p.m. on _____.
3. Please review the Chapter Objectives as listed on the first page of every chapter in the workbook and on your weekly syllabi's viewable on my web page at www.vcstulsa.org. They are also listed below by Chapter.
4. Please review the Chapter Key Terms as listed on the first page of every chapter in the workbook and on your weekly syllabi's viewable on my web page at www.vcstulsa.org. They are also listed below by Chapter.

CHAPTER OBJECTIVES (with cross-reference Workbook Pages notated)

UNIT 1 – SAVING and INVESTING

CHAPTER 1 – Savings (Pg. 9)

1. List Dave Ramsey's "Baby Steps" covered in *Foundations for Personal Finance*. (Pgs. 11, 12, 13, 33, 39, 59,171)
2. Explain the 3 basic reasons for saving money. (Pg. 12)
3. Identify the benefits of having an Emergency Fund. (Pg. 13)
4. Demonstrate how compound interest works and understand the impact of rate of return. (Pgs. 15, 16, 17)

CHAPTER 2 – Investment Options (Pg. 21)

1. Explain the KISS rule of investing. (Pg. 23)
2. Examine the relationship between diversification and risk. (Pgs. 23, 27)
3. Compare and contrast different types of investments including: money markets, bonds, single stocks, mutual funds, rental real estate, and annuities. (Pgs. 24, 25,26)

CHAPTER 3 – Wealth Building and College Savings (Pg. 31)

1. Explain what is meant by tax-favored dollars: Money that is working for you, either tax-deferred or tax free, within a retirement plan. (Pg. 33)
2. List the different types of retirement plans: Roth IRA, Traditional IRA, SEPP, 401(k), 403(b), 457. (Pgs. 34, 35, 36, 37)
3. Differentiate between a Traditional and Roth IRA. (Pgs. 34, 35, 38)
4. Illustrate how a 401(k) company match works and prioritizes money into various investments. (Pg. 37)
5. Describe how pre-tax and after-tax savings work. (Pgs. 39, 40)

UNIT 2 – CREDIT and DEBT

CHAPTER 4 – Dangers of Debt (Pg. 47)

1. Analyze the history and evolution of credit. (Pgs. 49, 50)
2. Demonstrate various ways people get into debt. (Pgs. 51, 52, 53, 54, 55, 56,57, 58, 59)
3. Compare and contrast credit cards and debit cards. (Pg. 57)
4. Explain how the debt snowball works. (Pg. 60)
5. Evaluate and refute the myths associated with debt. (Pgs. 51, 52, 53, 54, 55, 56, 57, 58, 59)

CHAPTER 5 – Consumer Awareness (Pg. 65)

1. List the ways companies compete for your money. (Pg. 67)
2. Evaluate the role "opportunity cost" plays in purchase decisions. (Pg. 69)
3. Distinguish what constitutes a significant purchase. (Pg. 68)
4. List the five steps to help you make a significant purchase buying decision. (Pg. 69)

CHAPTER 6 – Credit Bureaus and Collection Practices (Pg. 73)

1. Evaluate the myth of building your credit score. (Pg. 75)
2. Describe precautions that will protect you from identity theft. (Pg. 77)
3. Explain how to correct inaccuracies on a credit report. (Pg. 76, 77)
4. Become familiar with the Federal Fair Debt Collection Practices Act and parameters regulating collectors. (Pgs. 80,81)

Foundations In Personal Finance Packet Key Chapter 11

Michael Seilmaier



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web what mass of CoCl_2 in grams is needed for the solution express the answer using 3 significant figures

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web the hardness of water hardness count is usually expressed in parts per million by mass of CaCO_3 which is equivalent to milligrams of CaCO_3 per liter of water what is the molar concentration of Ca^{2+} ions in a water sample with a hardness count of 175 mg CaCO_3 l

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web answer the mass percentage of an element of a solution is defined as the mass of the solute in grammes present in 100 g of the solution it s written like this
$$\text{begin array l} \frac{\text{mass of a component}}{\text{mass of component in solution total mass of solution}} \times 100 \text{ end array}$$

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web aug 13 2017 molarity practice problems this chemistry video tutorial explains how to solve common molarity problems it discusses how to calculate the concentration of a solution given the mass in grams