## **Balancing Chemical Equations Practice**

Balance the following chemical equations.

1. Fe + 
$$O_2 \rightarrow$$
 Fe<sub>3</sub>O<sub>4</sub>

3. 
$$Sn + NaOH \rightarrow Na_2SnO_2 + H_2$$

4. 
$$K + Br_2 \rightarrow KBr$$

5. 
$$C_8H_{18} + O_2 \rightarrow CO_2 + H_2O$$

7. 
$$COCl_2 + H_2O \rightarrow HCl + CO_2$$

8. 
$$CS_2 + O_2 \rightarrow CO_2 + SO_2$$

9. 
$$H_2SO_4 + NaCN \rightarrow HCN + Na_2SO_4$$

11. 
$$H_2 + F_2 \rightarrow HF$$

12. BaCl<sub>2</sub> + KIO<sub>3</sub> 
$$\rightarrow$$
 Ba(IO<sub>3</sub>)<sub>2</sub> + KCl

13. 
$$Mg + HCl \rightarrow MgCl_2 + H_2$$

# **Naming And Balancing Chemical Equations Practice**

Pamela Mills, Amina El-Ashmawy

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Introductory Chemistry Neil Elbridge Gordon, 1927

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