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Assignment #3 – Compound Names and Formulas Single and Multi-valent ions

A. Write the correct formula for the following comp	oounds.		
1. gallium oxide <u>G0.203</u>	11. silicon oxide SiO2		
2. mercury (I) chloride HQCI	12. aluminum bromideA\BY3		
3. tin (IV) bromide Sn B/4	13. tin (II) sulphide SnS		
4. indium oxide <u>In205</u>	14. germanium bromide <u>Gr.BY4</u>		
5. cadmium sulphide <u>GdS</u>	15. nickel (II) oxide NiO		
6. magnesium oxide MOD	16. strontium astatide SY At 2		
7. potassium nitride K3N	17. aluminum oxide \A\2\3		
8. iron (III) sulphide Fc2S3	18. copper (II) oxide		
9. gold (III) bromide ANBY3	19. lawrencium fluoride LVF3		
10. zinc oxide ZY()	20. actinium oxide \(\frac{AC2()_3}{}		
B. Name the following compounds correctly. Only include the Roman Numeral in the name if the first element has 2 or more combining capacities.			
1. CaBr 2 COICIUM Bramide	9. AscloArsonic (V) Chloride		
2. K20 Potassium Oxide	10. EuzO3 Europium (II) Oxide		
3. Fe2O3 Iron (III) Oxide	11. NdBra Necdymium Bramide		
4. ZnF2 Zinc Fluoride	12. CusN2 Copper (II) NHride		
5. Cuci Copper (I) Chloride	13. Cos (MOOH (II) Sulphide		
6. NIZO3 NICKEL (III) Oxide	14. Zroz Zirconium Oxide		
7. Auci _ Gold (I) Chloride	15. Lizo Athium Oxide		
8. HgBr <u>HCYCUYY</u> (I) Bromide	16. NaH Sodium Hydride		

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Yan Bai

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