

## MATHEMATICAL STUDIES TZ1

### Overall grade boundaries

#### Standard level

<b>Grade:</b>	1	2	3	4	5	6	7
<b>Mark range:</b>	0 - 17	18 - 33	34 - 42	43 - 54	55 - 66	67 - 77	78 - 100

### Time zone variants of examination papers

To protect the integrity of the examinations, increasing use is being made of time zone variants of examination papers. By using variants of the same examination paper candidates in one part of the world will not always be taking the same examination paper as candidates in other parts of the world. A rigorous process is applied to ensure that the papers are comparable in terms of difficulty and syllabus coverage, and measures are taken to guarantee that the same grading standards are applied to candidates' scripts for the different versions of the examination papers. For the May 2014 examination session the IB has produced time zone variants of Mathematical Studies papers.

### Standard level Project

#### Component grade boundaries

<b>Grade:</b>	1	2	3	4	5	6	7
<b>Mark range:</b>	0 - 4	5 - 6	7 - 8	9 - 11	12 - 14	15 - 16	17 - 20

### The range and suitability of the work submitted

There was a wide range of marks as usual. Most of the topics were statistical and were suitable for a Mathematical Studies SL project but there are always a few that should have been actively discouraged by the teachers to start with. Some candidates had obviously worked hard on their project and enjoyed the process and this was obvious from the care that was taken to satisfy all the assessment criteria and, as a result, these candidates scored highly on all the criteria. However, there were others that showed little, if any, commitment and produced a trivial or incomplete piece of work. Some schools did not realize that the projects had to include two simple processes first before a further process was attempted and these scored poorly on Criterion C. Many candidates lost a mark due to improper notation and/or terminology or failing to define variables and teachers should

# Maths Studies 2paper Tz1

**RJ Shavelson**



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web elements in the second row of the periodic table place their electrons in the  $2n$  shell as well as the  $1n$  shell for instance lithium li has three electrons two fill the  $1s$  orbital and the third is placed in the  $2s$  orbital giving an electron configuration of  $1s^2 2s^1$

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web the electron removed during the ionization of beryllium he 2s 2 is an s electron whereas the electron removed during the ionization of boron he 2s 2 2p 1 is a p electron this results in a lower first ionization energy for boron even though its nuclear charge is greater by one proton

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web predictions from reliable sources have been used for these elements grayed out electron numbers indicate subshells filled to their maximum bracketed noble gas symbols on the left represent inner configurations that are the same in each period written out these are he 2 helium 1s 2 ne 10 neon 1s 2 2s 2 2p 6

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