

PHYSICAL: Kinetic Molecular Theory

Learning Objectives

- At standard atmospheric conditions, all gases have approximately the same kinetic energy, and are made up of molecules that behave as independent particles (the particles of the gases are relatively small).
- At standard atmospheric conditions, all gases have the same average kinetic energy.
- At standard atmospheric conditions, all gases have the same average molecular mass.

Model 1: Representation of Gases in Different Phases

The following figure illustrates the representation of particles in three different phases: solid, liquid, and gas.



Figure 1: Representation of particles in three different phases: solid, liquid, and gas.

At standard atmospheric conditions, all gases have the same average kinetic energy, and are made up of molecules that behave as independent particles (the particles of the gases are relatively small).

Key Concepts

- At standard atmospheric conditions, all gases have the same average kinetic energy, and are made up of molecules that behave as independent particles (the particles of the gases are relatively small).
- At standard atmospheric conditions, all gases have the same average molecular mass.
- At standard atmospheric conditions, all gases have the same average molecular mass.

Table 1: Representation of particles in three different phases: solid, liquid, and gas.

	SOLID	LIQUID	GAS
How are the particles arranged?	Particles are packed closely together in a regular, repeating pattern. The particles are held together by strong forces.	Particles are packed closely together in a disordered, repeating pattern. The particles are held together by moderate forces.	Particles are widely spaced and move freely. The particles are held together by very weak forces.
How do the particles move?	Particles vibrate in place. They have a fixed position and can only move in a limited range of motion.	Particles move in a disordered, repeating pattern. They have a fixed position and can move in a limited range of motion.	Particles move in a disordered, repeating pattern. They have a fixed position and can move in a limited range of motion.

Lw Pogil Kinetic Molecular Theory Key

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The Enigmatic Realm of **Lw Pogil Kinetic Molecular Theory Key**: Unleashing the Language is Inner Magic

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