

## Ch 13 Study Guide (Chemical Bonds)

1. Fill in the table below with the correct values:

Element Name	Ion Symbol	# Protons	# Electrons	Charge	Ion Type
Fluorine	$F^-$	9	10	-1	Anion
Iodine	$I^-$	53	54	-1	Anion
Sulfur	$S^{2-}$	16	18	-2	Anion
Potassium	$K^+$	19	18	+1	Cation
Calcium	$Ca^{2+}$	20	18	+2	Cation
Bromine	$Br^-$	35	36	-1	Anion
Strontium	$Sr^{2+}$	38	36	+2	Cation
Oxygen	$O^{2-}$	8	10	-2	Anion
Magnesium	$Mg^{2+}$	12	10	+2	Cation
Aluminum	$Al^{3+}$	13	10	+3	Cation
Selenium	$Se^{2-}$	34	36	-2	Anion
Lithium	$Li^+$	3	2	+1	Cation
Rubidium	$Rb^+$	37	36	+1	Cation
Chlorine	$Cl^-$	17	18	-1	Anion

2. An ionic bond is formed between what two types of atoms?

Metal & nonmetal

3. A covalent bond is formed between what two types of atoms?

Nonmetal & Nonmetal

4. Draw the bond(s) that will form between Mg and O.



5. What is the formula for the compound formed in question #4?



# Ionic And Metallic Bonding Chapter Test B Answer Key

**Rob Ritchie**



## **Ionic And Metallic Bonding Chapter Test B Answer Key:**

## Unveiling the Magic of Words: A Review of "**Ionic And Metallic Bonding Chapter Test B Answer Key**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Ionic And Metallic Bonding Chapter Test B Answer Key**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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