

- Directions:**
 1. An element may be assigned a positive or negative charge depending on whether it is a metal or nonmetal.
 2. An atom with a positive charge is a cation.
 3. An atom with a negative charge is an anion.
 4. The number of protons in an atom is the atomic number.

Ion Practice Set

1. Write the symbol for the ion that has the following number of protons and electrons:
 18 protons, 19 electrons
2. Write the symbol for the ion that has the following number of protons and electrons:
 19 protons, 18 electrons

Complete the following table, using the periodic table in the back of your book.

	ELEMENT NAME	ION SYMBOL	NUMBER OF PROTONS	NUMBER OF ELECTRONS	CHARGE OF ION (POSITIVE OR NEGATIVE)	TYPE OF ION
1	Sodium	Na^+	11	10	positive	cation
2	Chlorine	Cl^-	17	18	negative	anion
3	Sulfur	S^{2-}	16	18	negative	anion
4	Calcium	Ca^{2+}	20	18	positive	cation
5	Fluorine	F^-	9	10	negative	anion
6	Magnesium	Mg^{2+}	12	10	positive	cation
7	Hydrogen	H^+	1	0	positive	cation
8	Oxygen	O^{2-}	8	10	negative	anion
9	Aluminum	Al^{3+}	13	10	positive	cation
10	Strontium	Sr^{2+}	38	36	positive	cation
11	Selenium	Se^{2-}	34	36	negative	anion
12	Barium	Ba^{2+}	56	54	positive	cation
13	Iron	Fe^{2+}	26	24	positive	cation
14	Gold	Au^+	79	78	positive	cation
15	Carbon	C^{4-}	6	10	negative	anion

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G Thomas

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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