

## Inorganic Chemistry 2 Exam Review With 100% Correct Answers 2024

1) Which of the following statements about gases is false?

A) All gases are colorless and odorless at room temperature.

B) Distances between molecules of gas are very large compared to bond distances within molecules.

C) Gases expand spontaneously to fill the container they are placed in.

D) Gases are highly compressible.

E) Non-reacting gas mixtures are homogeneous. - answer ✓✓ A) All gases are colorless and odorless at room temperature.

2) The dissolution of water in octane ( $C_8H_{18}$ ) is principally prevented by \_\_\_\_\_.

A) dipole-dipole attraction between octane molecules

B) repulsion between like -charged water and octane molecules

C) hydrogen bonding between water molecules

D) London dispersion forces between octane molecules

# Inorganic Chemistry Practice Exam Answers

**Judd E. Hollander**



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