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$$m = \frac{\text{moles of solute}}{\text{kg of solvent}}$$

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$$m_{mix} = \frac{m_1 v_1 + m_2 v_2 + \dots + m_n v_n}{v_1 + v_2 + \dots + v_n}$$
 1 5 x 480 1 2 x 520 480 520 720 624 1000 1 344 m q5 with increase in temperature which of these changes a molality b fraction of solute present in water

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