

Chemistry for Health Science

Appendix A-B

- **Significant Figures:**

1. *A Number is a Sig Fig if it is:*

- Not a zero: 4.5g (2 sig fig), 122.35m (5 sig figs)
- A zero between nonzero digits: 205 (3 sig figs), 5.008kg (4 sig figs).
- A zero at the end of a decimal number: 50L (2 sig figs), 16.00 (4 sig figs).
- In the coefficient of a number written in scientific notation: 4.8×10^5 m (2 sig figs), 5.70×10^{-3} g (3 sig figs).

2. *A zero is not significant if it is:*

- At the beginning of a decimal number: 0.00045 (2 sig figs), 0.075 (2 sig figs).
- Used as a placeholder in a large number without decimal point: 850000m (2 sig figs), 1,250,000 (3 sig figs).

- **Rules to Round Numbers:**

- If the first to be dropped 4 or less, then it and all the following digits are dropped from the number.
- 5 or greater, then the last retained digit of the number is increased by 1.

Ex: 8.4234=8.42, 14,780=14.8, 32560= 3.26×10^5

Ex: 1. 3.145=3.15 2. 824.75cm=825cm 3. 0.112486g=0.112g 4. 5.3L=5.30L

- **Scientific Notation:**

- Used when you have a very big number or a very small one.
- The coefficient is 1 but less than 10.

Ex: 2,400= 2.4×10^3 , 0.00086= 8.6×10^{-4}

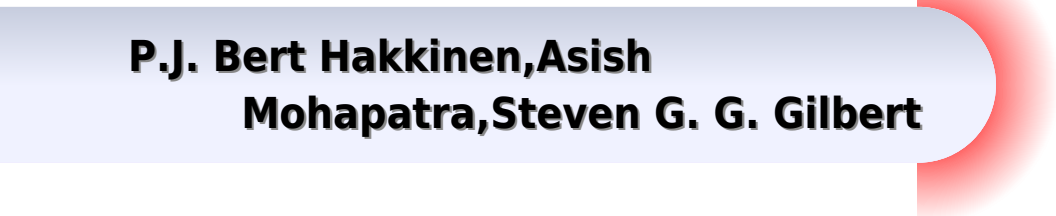
Ex: 1. 64000= 6.4×10^4 2. 0.021= 2.1×10^{-2} 3. 0.000008= 8×10^{-6}

- **Decimal Places Subtract/ Add**

- Use the number with the fewer decimal place in the final result.

Health Science Kids Study Guide

**P.J. Bert Hakkinen, Asish
Mohapatra, Steven G. G. Gilbert**



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