



- 3.5 Substance exists as both a solid and a liquid ✓✓ or... is melting (2)
 3.6 37 °C ✓ (2)
 3.7 80 °C ✓ (1)
 3.8 An indication of the average kinetic energy ✓ of the particles in a substance. ✓ (2)
 3.9 Physical ✓ (1)
 3.10 In the solid state the intermolecular forces are the strongest and the particles cannot move. The heater provides thermal energy to the system, which is translated into kinetic energy (increase in temperature) until the intermolecular forces between the particles can be broken. ✓✓ At this point the thermal energy is transformed into latent energy to break the bonds, so no increase in temperature occurs. ✓✓ (4)

[23]

QUESTION 4

- 4.1
 4.1.1 Homogeneous ✓ (1)
 4.1.2 Homogeneous ✓ (1)
 4.1.3 Pure substance ✓ (1)
 4.2
 4.2.1 Potassium bromide ✓ (2)
 4.2.2 Calcium carbonate ✓ (2)
 4.2.3 Ammonium chloride ✓ (2)
 4.3
 4.3.1 ZnSO_4 ✓✓ (2)
 4.3.2 KMnO_4 ✓✓ (2)
 4.3.3 H_2O ✓✓ (2)

[17]

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 - Advantages of eBooks Over Traditional Books
2. Identifying Grade 1life Science Question Paper Of March 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
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 - Popular eBook Platforms
 - Features to Look for in an Grade 1life Science Question Paper Of March 2014
 - User-Friendly Interface
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 - Personalized Recommendations
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 - Grade 1life Science Question Paper Of March 2014 and Bestseller Lists
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 - Grade 1life Science Question Paper Of March 2014 Budget-Friendly Options
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 - Grade 1life Science Question Paper Of March 2014 Compatibility with Devices
 - Grade 1life Science Question Paper Of March 2014 Enhanced eBook Features
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- Adjustable Fonts and Text Sizes of Grade 1life Science Question Paper Of March 2014
 - Highlighting and Note-Taking Grade 1life Science Question Paper Of March 2014
 - Interactive Elements Grade 1life Science Question Paper Of March 2014
8. Staying Engaged with Grade 1life Science Question Paper Of March 2014
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 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Grade 1life Science Question Paper Of March 2014
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- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
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- Setting Reading Goals Grade 1life Science Question Paper Of March 2014
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Grade 1life Science Question Paper Of March 2014
- Fact-Checking eBook Content of Grade 1life Science Question Paper Of March 2014
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
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