



Mazda Bravo B2600 Workshop Manual

Michael Hoelscher



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web the relative electronegativities of the bonded atoms is $\text{H} < \text{C} < \text{Cl}$ and so the bond moments all point toward the Cl end of the
 molecule and sum to yield a considerable dipole moment the molecules are relatively polar for molecules of high symmetry
 such as BF_3 trigonal planar CH_4 tetrahedral PF_5 trigonal bipyramidal and SF_6

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