

# MULTI-STEP PROBLEMS: CHAPTER 25

## Problem 25.4

The C-Cl bonds of the two chloroethoxyethyl acetate derivatives are characterized by peaks at 840 and 1040 cm<sup>-1</sup> (trans-1,2-dichloro-1,2-dichloroethoxyethyl acetate) and 840 and 1040 cm<sup>-1</sup> (trans-1,2-dichloro-1,2-dichloroethoxyethyl acetate). The peaks at 840 and 1040 cm<sup>-1</sup> are assigned to the C-Cl stretching vibrations. The peaks at 840 and 1040 cm<sup>-1</sup> are assigned to the C-Cl stretching vibrations. The peaks at 840 and 1040 cm<sup>-1</sup> are assigned to the C-Cl stretching vibrations.

## Part a

Sketch the electric dipole moment of the C-Cl bond (also represented as a vector)  $\vec{\mu}_{C-Cl}$  and  $\vec{\mu}_{H-Cl}$ .

$$\vec{\mu} = \frac{q \cdot r}{4\pi\epsilon_0}$$

$$\vec{\mu}_{C-Cl} = \frac{q_{Cl} \cdot r_{C-Cl}}{4\pi\epsilon_0} = \frac{(1.6 \times 10^{-19} \text{ C}) \cdot (1.7 \times 10^{-10} \text{ m})}{4\pi\epsilon_0}$$

$$\vec{\mu}_{H-Cl} = \frac{q_{Cl} \cdot r_{H-Cl}}{4\pi\epsilon_0} = \frac{(1.6 \times 10^{-19} \text{ C}) \cdot (1.3 \times 10^{-10} \text{ m})}{4\pi\epsilon_0}$$

$$\vec{\mu}_{C-Cl} = \frac{(1.6 \times 10^{-19} \text{ C}) \cdot (1.7 \times 10^{-10} \text{ m})}{4\pi\epsilon_0} = 1.7 \times 10^{-29} \text{ C} \cdot \text{m}$$

$$\vec{\mu}_{H-Cl} = \frac{(1.6 \times 10^{-19} \text{ C}) \cdot (1.3 \times 10^{-10} \text{ m})}{4\pi\epsilon_0} = 1.3 \times 10^{-29} \text{ C} \cdot \text{m}$$

$$\vec{\mu}_{C-Cl} = \frac{q_{Cl} \cdot r_{C-Cl}}{4\pi\epsilon_0} = \frac{(1.6 \times 10^{-19} \text{ C}) \cdot (1.7 \times 10^{-10} \text{ m})}{4\pi\epsilon_0}$$

$$\vec{\mu}_{H-Cl} = \frac{q_{Cl} \cdot r_{H-Cl}}{4\pi\epsilon_0} = \frac{(1.6 \times 10^{-19} \text{ C}) \cdot (1.3 \times 10^{-10} \text{ m})}{4\pi\epsilon_0}$$

$$\vec{\mu}_{C-Cl} = \frac{(1.6 \times 10^{-19} \text{ C}) \cdot (1.7 \times 10^{-10} \text{ m})}{4\pi\epsilon_0} = 1.7 \times 10^{-29} \text{ C} \cdot \text{m}$$

$$\vec{\mu} = \vec{\mu}_{C-Cl} + \vec{\mu}_{H-Cl} = (1.7 \times 10^{-29} + 1.3 \times 10^{-29}) \text{ C} \cdot \text{m} = \boxed{3.0 \times 10^{-29} \text{ C} \cdot \text{m}}$$

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